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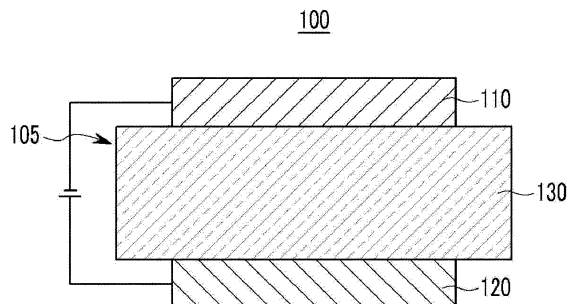
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(54) **COMPOUND FOR ORGANIC OPTOELECTRIC DEVICE, ORGANIC LIGHT-EMITTING DIODE INCLUDING SAME, DISPLAY DEVICE INCLUDING ORGANIC LIGHT-EMITTING DIODE**

(57) Provided are a compound for an organic optoelectric device represented by Chemical Formula 1, an organic light emitting diode including the same, and a display device including the organic light emitting diode. The structure of the compound for an organic optoelectric device represented by Chemical Formula 1 is described in the specification.

The compound for an organic photoelectric device provides an organic light emitting diode having excellent electrochemical and thermal stability and improved life-span characteristics, and high luminous efficiency at a low driving voltage, and the compound for an organic photoelectric device may be appropriate for a solution process.

【Figure 1】



Description**[Invention Title]**

- 5 **[0001]** Compound for Organic Optoelectric Device, Organic Light Emitting Diode and Display Device Including Organic Light Emitting Diode

[Technical Field]

- 10 **[0002]** A compound for an organic optoelectric device, an organic light emitting diode including the same, and a display device including the organic light emitting diode are disclosed.

[Background Art]

- 15 **[0003]** An organic optoelectric device is a device requiring a charge exchange between an electrode and an organic material by using holes or electrons.

- [0004]** An organic optoelectric device may be classified as follows in accordance with its driving principles. A first organic optoelectric device is an electronic device driven as follows: excitons are generated in an organic material layer by photons from an external light source; the excitons are separated into electrons and holes; and the electrons and holes are transferred to different electrodes as a current source (voltage source).

- 20 **[0005]** A second organic optoelectric device is an electronic device driven as follows: a voltage or a current is applied to at least two electrodes to inject holes and/or electrons into an organic material semiconductor positioned at an interface of the electrodes, and the device is driven by the injected electrons and holes.

- [0006]** Examples of the organic optoelectric device includes organic photoelectric device, an organic light emitting diode, an organic solar cell, an organic photo conductor drum, and an organic transistor, and the like, which requires a hole injecting or transport material, an electron injecting or transport material, or a light emitting material.

- 25 **[0007]** Particularly, an organic light emitting diode (OLED) has recently drawn attention due to an increasing demand for a flat panel display. In general, organic light emission refers to conversion of electrical energy into photo-energy.

- [0008]** Such an organic light emitting diode converts electrical energy into light by applying current to an organic light emitting material. It has a structure in which a functional organic material layer is interposed between an anode and a cathode. The organic material layer includes a multi-layer including different materials, for example a hole injection layer (HIL), a hole transport layer (HTL), an emission layer, an electron transport layer (ETL), and an electron injection layer (EIL), in order to improve efficiency and stability of an organic light emitting diode.

- 30 **[0009]** In such an organic light emitting diode, when a voltage is applied between an anode and a cathode, holes from the anode and electrons from the cathode are injected to an organic material layer and recombined to generate excitons having high energy. The generated excitons generate light having certain wavelengths while shifting to a ground state.

- [0010]** Recently, it has become known that a phosphorescent light emitting material can be used for a light emitting material of an organic light emitting diode in addition to the fluorescent light emitting material. Such a phosphorescent material emits lights by transporting the electrons from a ground state to an excited state, non-radiance transiting of a singlet exciton to a triplet exciton through intersystem crossing, and transiting a triplet exciton to a ground state to emit light.

- 40 **[0011]** As described above, in an organic light emitting diode, an organic material layer includes a light emitting material and a charge transport material, for example a hole injection material, a hole transport material, an electron transport material, an electron injection material, and the like.

- [0012]** The light emitting material is classified as blue, green, and red light emitting materials according to emitted colors, and yellow and orange light emitting materials to emit colors approaching natural colors.

- 45 **[0013]** When one material is used as a light emitting material, a maximum light emitting wavelength is shifted to a long wavelength or color purity decreases because of interactions between molecules, or device efficiency decreases because of a light emitting quenching effect, and therefore, a host/dopant system is included as a light emitting material in order to improve color purity and increase luminous efficiency and stability through energy transfer.

- 50 **[0014]** In order to implement excellent performance of an organic light emitting diode, a material constituting an organic material layer, for example a hole injection material, a hole transport material, a light emitting material, an electron transport material, an electron injection material, and a light emitting material such as a host and/or a dopant, should be stable and have good efficiency. However, development of an organic material layer forming material for an organic light emitting diode has thus far not been satisfactory and thus there is a need for a novel material. This material development is also required for other organic optoelectric devices.

- 55 **[0015]** The low molecular organic light emitting diode is manufactured as a thin film in a vacuum deposition method and can have good efficiency and life-span performance. A polymer organic light emitting diode is manufactured in an Inkjet or spin coating method has an advantage of low initial cost and being large-sized.

[0016] Both low molecular organic light emitting and polymer organic light emitting diodes have an advantage of self-light emitting, high speed response, wide viewing angle, ultra-thin, high image quality, durability, large driving temperature range, and the like. In particular, they have good visibility due to self-light emitting characteristic compared with a conventional LCD (liquid crystal display) and have an advantage of decreasing thickness and weight of LCD up to a third, because they do not need a backlight.

[0017] In addition, since they have a response speed of a microsecond unit, which is 1000 time faster than LCD, they can realize a perfect motion picture without after-image. Based on these advantages, they have been remarkably developed to have 80 times efficiency and more than 100 times life-span since they come out for the first time in the late 1980s and recently, they keep being rapidly larger such as a 40-inch organic light emitting diode panel.

[0018] They are simultaneously required to have improved luminous efficiency and life-span in order to be larger. Therefore, a stable and efficient organic material layer material for an organic light emitting diode needs to be developed.

[DISCLOSURE]

[Technical Problem]

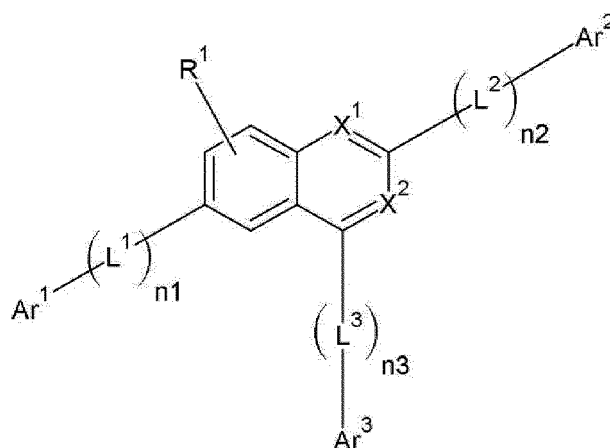
[0019] One embodiment provides a compound being capable of providing an organic optoelectric device having high efficiency and long life-span.

[0020] Another embodiment provides an organic light emitting diode including the compound and a display device including the organic light emitting diode.

[Technical Solution]

[0021] In one embodiment of the present invention, a compound represented by Chemical Formula 1 for an organic optoelectric device is provided.

[Chemical Formula 1]



[0022] In Chemical Formula 1, X¹ and X² are each independently N or CR', at least one of X¹ and X² is N, Ar³ is a substituted or unsubstituted C2 to C30 heteroaryl group, Ar¹ and Ar² are each independently a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C2 to C30 heteroaryl group, L¹ to L³ are each independently a substituted or unsubstituted C6 to C30 arylene group or a substituted or unsubstituted C2 to C30 heteroarylene group, n1 to n3 are 0 or 1,

R¹ and R' are each independently hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

[0023] In another embodiment of the present invention, provided is an organic light emitting diode that includes an anode, a cathode and at least one organic thin layer between the anode and the cathode, wherein at least one layer of the organic thin layer includes the compound according to the embodiment of the present invention.

[0024] In yet another embodiment of the present invention, a display device including the organic light emitting diode according to the embodiment of the present invention is provided.

[Advantageous Effects]

[0025] An organic optoelectric device including the compound according to the embodiment of the present invention has excellent electrochemical and thermal stability, improved life-span characteristics, and high luminous efficiency at a low driving voltage. In addition, the compound may be appropriate for a solution process.

[Description of the Drawings]

[0026] FIGS. 1 and 2 are cross-sectional views showing various embodiments of organic light emitting diodes according to embodiments of the present invention.

<Description of Reference Numerals Indicating Primary Elements in the Drawings>

[0027]

100, 200: organic light emitting diode	105: organic layer
110: cathode	120: anode
130, 230: emission layer	140: hole auxiliary layer

[Mode for Invention]

[0028] Hereinafter, embodiments of the present invention are described in detail. However, these embodiments are exemplary, the present invention is not limited thereto and the present invention is defined by the scope of claims.

[0029] In the present specification, when a definition is not otherwise provided, the term "substituted" refers to one substituted with a substituent selected from deuterium, a halogen, hydroxy group, an amino group, a substituted or unsubstituted C1 to C30 amine group, a nitro group, a substituted or unsubstituted C1 to C40 silyl group, C1 to C30 alkyl group, C1 to C10 alkylsilyl group, C3 to C30 cycloalkyl group, C6 to C30 aryl group, C1 to C20 alkoxy group, a fluoro group, a C1 to C10 trifluoroalkyl group such as a trifluoromethyl group or cyano group instead of at least one hydrogen of a functional group.

[0030] In addition, two adjacent substituents of the substituted halogen, hydroxy group, amino group, substituted or unsubstituted C1 to C20 amine group, nitro group, substituted or unsubstituted C3 to C40 silyl group, C1 to C30 alkyl group, C1 to C10 alkylsilyl group, C3 to C30 cycloalkyl group, C6 to C30 aryl group, C1 to C20 alkoxy group, fluoro group, C1 to C10 trifluoroalkyl group such as trifluoromethyl group and the like, or cyano group may be fused with each other to form a ring.

[0031] In the present specification, when specific definition is not otherwise provided, the term "hetero" refers to one including 1 to 3 hetero atoms selected from N, O, S, and P, and remaining carbons in one compound or substituent.

[0032] In the present specification, when a definition is not otherwise provided, the term "combination thereof" refers to at least two substituents bound to each other by a linker, or at least two substituents condensed to each other.

[0033] In the present specification, when a definition is not otherwise provided, "alkyl group" refers to an aliphatic hydrocarbon group. The alkyl group may be "a saturated alkyl group" without any double bond or triple bond.

[0034] The alkyl group may be a C1 to C20 alkyl group. More specifically, the alkyl group may be a C1 to C10 alkyl group or a C1 to C6 alkyl group. For example, a C1 to C4 alkyl group may have 1 to 4 carbon atoms in an alkyl chain which may be selected from methyl, ethyl, propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl, and t-butyl.

[0035] Specific examples of the alkyl group may be a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a t-butyl group, a pentyl group, a hexyl group, a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, and the like.

[0036] In the present specification, "aryl group" refers to a cyclic substituent where all elements have p-orbitals, and these p-orbitals forms conjugation, and includes a monocyclic or fused ring polycyclic (i.e., rings sharing adjacent pairs of carbon atoms) functional group.

[0037] In the present specification, "heteroaryl group" refers to aryl group including 1 to 3 hetero atoms selected from N, O, S, and P, and remaining carbons. When the heteroaryl group is a fused ring, each ring may include 1 to 3 hetero atoms.

[0038] More specifically, the substituted or unsubstituted C6 to C30 aryl group and/or the substituted or unsubstituted C2 to C30 heteroaryl group refer to a substituted or unsubstituted phenyl group, a substituted or unsubstituted naphthyl group, a substituted or unsubstituted anthracenyl group, a substituted or unsubstituted phenanthryl group, a substituted

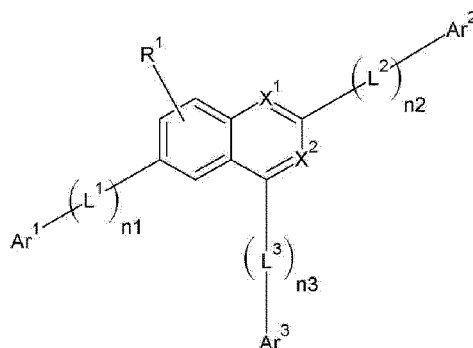
or unsubstituted naphthacenyl group, a substituted or unsubstituted pyrenyl group, a substituted or unsubstituted biphenyl group, a substituted or unsubstituted p-terphenyl group, a substituted or unsubstituted m-terphenyl group, a substituted or unsubstituted chrysenyl group, a substituted or unsubstituted triphenylenyl group, a substituted or unsubstituted perylenyl group, a substituted or unsubstituted indenyl group, a substituted or unsubstituted furanyl group, a substituted or unsubstituted thiophenyl group, a substituted or unsubstituted pyrrolyl group, a substituted or unsubstituted pyrazolyl group, a substituted or unsubstituted imidazolyl group, a substituted or unsubstituted triazolyl group, a substituted or unsubstituted oxazolyl group, a substituted or unsubstituted thiazolyl group, a substituted or unsubstituted oxadiazolyl group, a substituted or unsubstituted thiadiazolyl group, a substituted or unsubstituted fluorenyl group, a substituted or unsubstituted dibenzofuranyl group, a substituted or unsubstituted dibenzothiophenyl group, a substituted or unsubstituted carbazolyl group, a substituted or unsubstituted pyridyl group, a substituted or unsubstituted pyrimidinyl group, a substituted or unsubstituted pyrazinyl group, a substituted or unsubstituted triazinyl group, a substituted or unsubstituted benzofuranyl group, a substituted or unsubstituted benzothiophenyl group, a substituted or unsubstituted benzimidazolyl group, a substituted or unsubstituted indolyl group, a substituted or unsubstituted quinoliny group, a substituted or unsubstituted isoquinoliny group, a substituted or unsubstituted quinazoliny group, a substituted or unsubstituted quinoxaliny group, a substituted or unsubstituted naphthyridiny group, a substituted or unsubstituted benzoxazinyl group, a substituted or unsubstituted benzthiazinyl group, a substituted or unsubstituted acridinyl group, a substituted or unsubstituted phenazinyl group, a substituted or unsubstituted phenothiazinyl group, a substituted or unsubstituted phenoxazinyl group, or a combination thereof, but are not limited thereto.

[0039] In the present specification, hole characteristics refer to characteristics that holes formed in the anode is easily injected into the emission layer and transported in the emission layer due to conductive characteristics according to HOMO level. More specifically, it is similar to electron-repelling characteristics.

[0040] Electron characteristics refer to characteristics that electron formed in the cathode is easily injected into the emission layer and transported in the emission layer due to conductive characteristics according to LUMO level. More specifically, it is similar to electron-withdrawing characteristics.

[0041] In one embodiment of the present invention, a compound represented by Chemical Formula 1 for an organic optoelectric device is provided.

[Chemical Formula 1]



[0042] In Chemical Formula 1, X^1 and X^2 are each independently N or CR', at least one of X^1 and X^2 is N, Ar^3 is a substituted or unsubstituted C2 to C30 heteroaryl group, Ar^1 and Ar^2 are each independently a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C2 to C30 heteroaryl group, L^1 to L^3 are each independently a substituted or unsubstituted C6 to C30 arylene group or a substituted or unsubstituted C2 to C30 heteroarylene group, n_1 to n_3 are independently 0 or 1, and R^1 and R' are each independently hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group, or a substituted or unsubstituted silyl group.

[0043] The compound according to one embodiment of the present invention includes a substituted or unsubstituted C2 to C30 heteroaryl group at the Ar^3 , which gives a steric effect and may improve amorphous properties. In addition, charge transport characteristics of the compound may be improved.

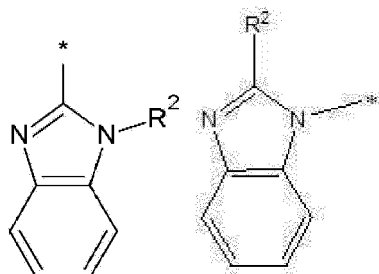
[0044] In addition, the compound represented by Chemical Formula 1 may have various energy bandgaps due to various substituents.

[0045] The compound may have an appropriate energy level depending on the substituents and thus, may fortify hole

transport capability or electron transport capability of an organic optoelectric device and bring about excellent effects on efficiency and driving voltage and also, have excellent electrochemical and thermal stability and thus, improve life-span characteristics during the operation of the organic optoelectric device.

[0046] More specifically, the Ar³ may be a substituent represented by Chemical Formula 2 or A.

[Chemical Formula 2] [Chemical Formula A]



[0047] In Chemical Formulae 2 and A, R² is hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

[0048] The substituent such as Chemical Formula 2 has a high glass transition temperature such as a high thermal decomposition temperature and thus improves life-span characteristics.

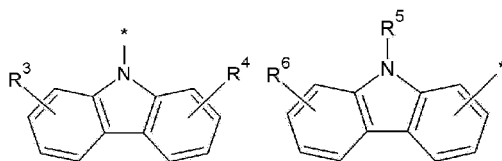
[0049] Or, the Ar³ may be a substituted or unsubstituted pyridinyl group, a substituted or unsubstituted pyrimidinyl group, a substituted or unsubstituted triazinyl group, a substituted or unsubstituted quinolinyl group, or a substituted or unsubstituted isoquinolinyl group. In this case, charge injection and charge mobility may be improved.

[0050] In one embodiment of the present invention, the Ar¹ and Ar² may be each independently a substituted or unsubstituted C6 to C30 aryl group. More specifically, the Ar¹ and Ar² may be a substituted or unsubstituted phenyl group, a substituted or unsubstituted biphenyl group, a substituted or unsubstituted naphthyl group, a substituted or unsubstituted anthracenyl group, or a substituted or unsubstituted phenanthrenyl group. In this case, hole and/or electron characteristics of the compound may be appropriately adjusted.

[0051] In one embodiment of the present invention, at least one of the Ar¹ and Ar² may be a substituted or unsubstituted carbazolyl group. In this case, charge mobility may be improved due to appropriate molecular energy and triplet energy, and the compound may have characteristics appropriate for an emission layer due to characteristics of the carbazolyl group.

[0052] For more specific examples, at least one of the Ar¹ and Ar² may be a substituent represented by Chemical Formula 3 or 4.

[Chemical Formula 3] [Chemical Formula 4]



[0053] In Chemical Formulae 3 and 4, R³ to R⁶ are independently hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

[0054] In one embodiment of the present invention, the L³ may be a substituted or unsubstituted C2 to C30 heteroarylene group. In this case, electron characteristics of the compound may be fortified, and thus charge mobility increases and high efficiency may be provided.

[0055] For more specific examples, the L³ may be a substituted or unsubstituted pyridinylene group, a substituted or unsubstituted pyrimidinylene group or a substituted or unsubstituted triazinylene group. But it is not limited thereto.

[0056] The L¹ to L³ may be selectively adjusted to determine an entire conjugation length of the compound, and a triplet energy bandgap of the compound may be adjusted therefrom. Thereby, characteristics of a material required of an organic optoelectric device may be realized. In addition, the triplet energy bandgap may also be adjusted by changing

a bonding position of ortho, para, and meta.

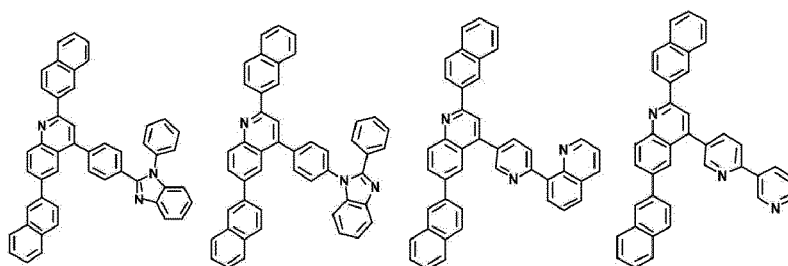
[0057] Simultaneously, the L^1 and L^2 may be a substituted or unsubstituted C6 to C30 arylene group, and in this case, the compound may have appropriate hole and electron characteristics.

[0058] Specific examples of the L^1 and L^2 may be a substituted or unsubstituted phenylene group, a substituted or unsubstituted biphenylene group, a substituted or unsubstituted terphenylene group, a substituted or unsubstituted naphthylene group, a substituted or unsubstituted anthracenylene group, a substituted or unsubstituted phenanthrylene group, a substituted or unsubstituted pyrenylene group, a substituted or unsubstituted fluorenylene group, a substituted or unsubstituted p-terphenyl group, a substituted or unsubstituted m-terphenyl group, a substituted or unsubstituted perylenyl group, and the like.

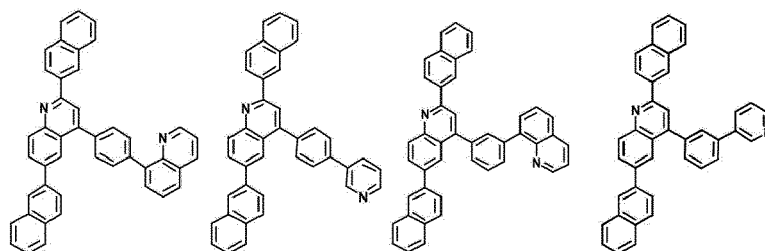
[0059] In one embodiment of the present invention, the X^1 may be N, and the X^2 may be CR'. But they are not limited thereto.

[0060] Specific examples of the compound according to one embodiment of the present invention are as follows, but are not limited thereto.

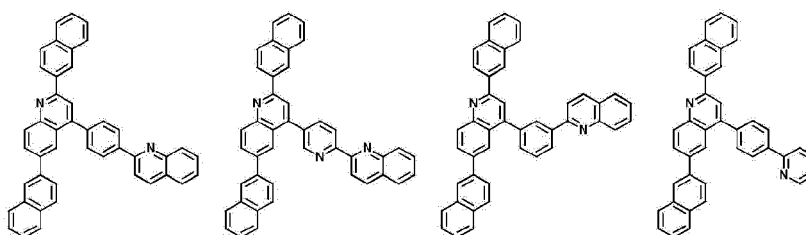
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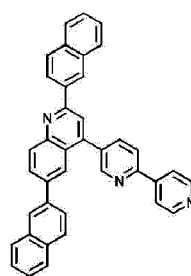
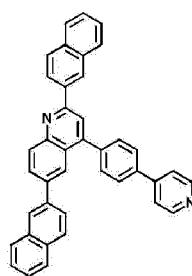
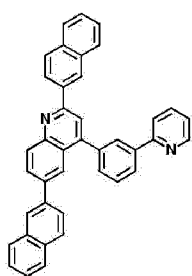
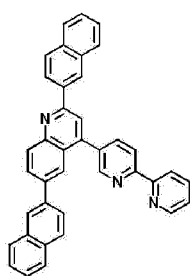
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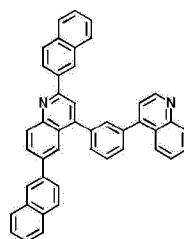
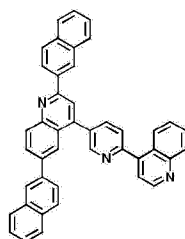
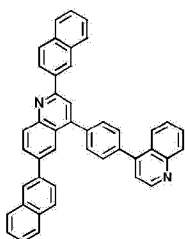
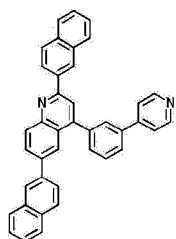
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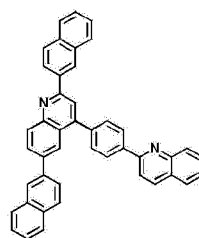
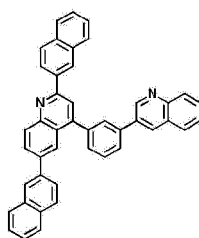
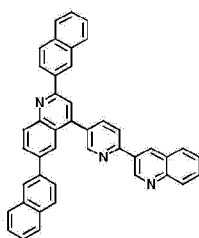
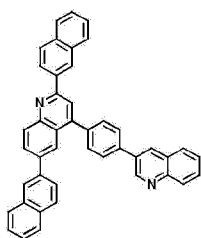
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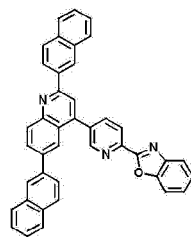
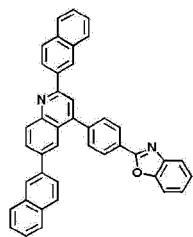
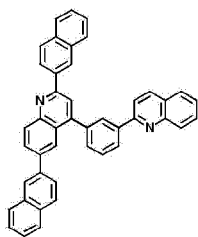
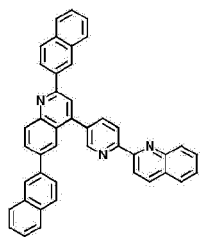
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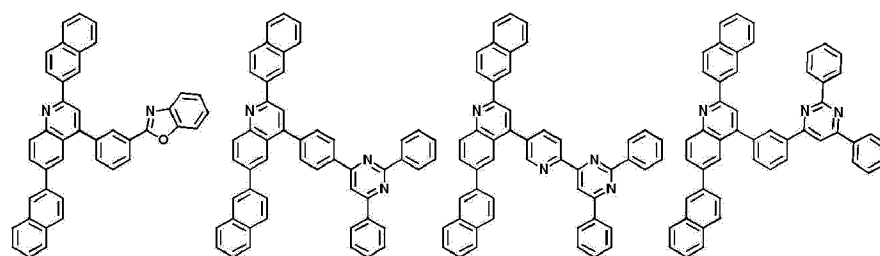
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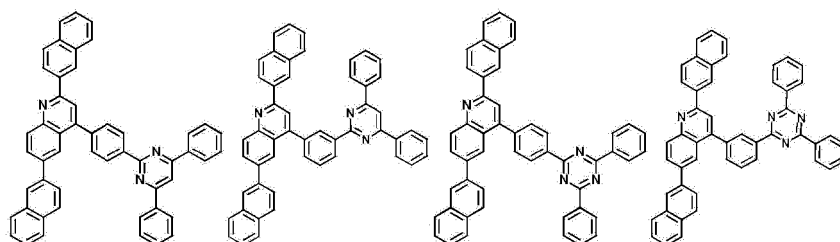
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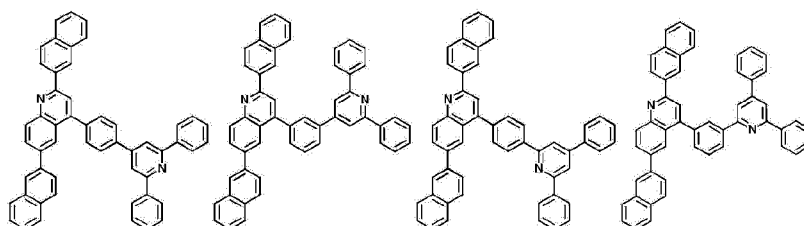
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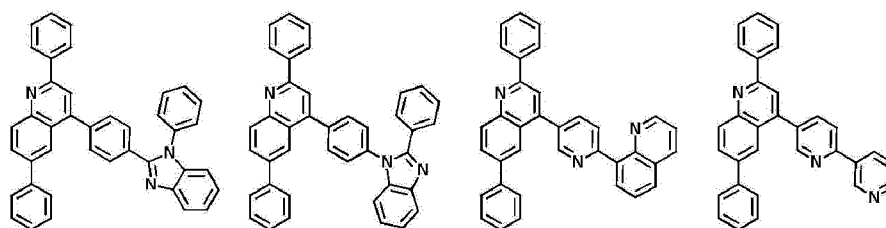
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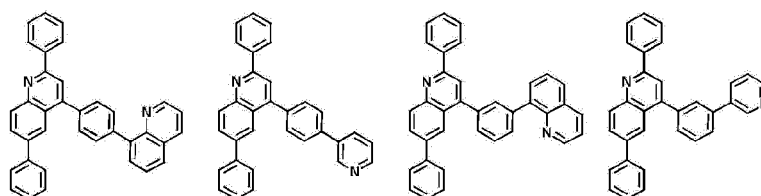
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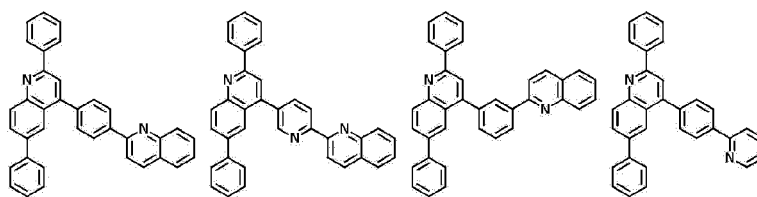


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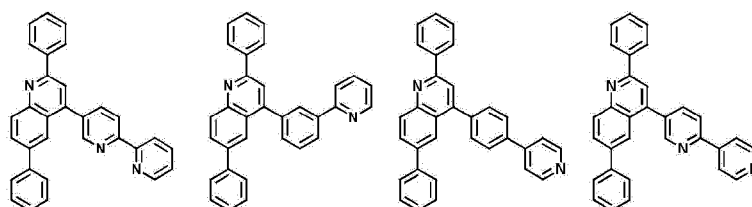
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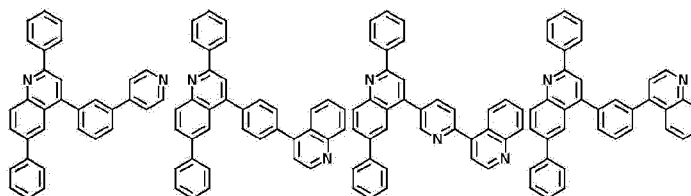
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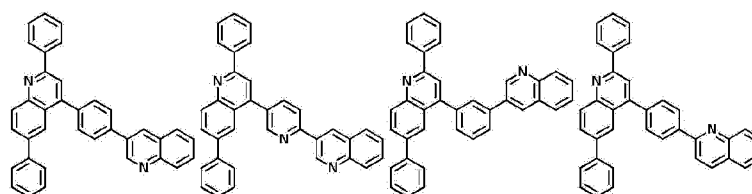
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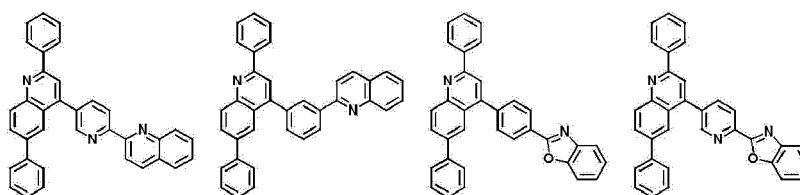
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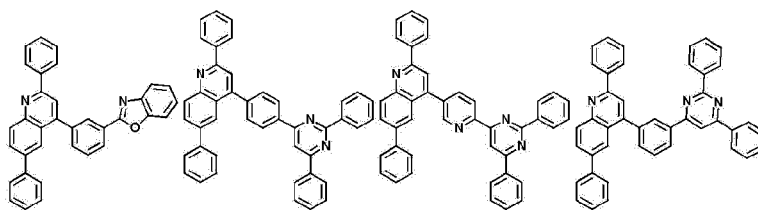


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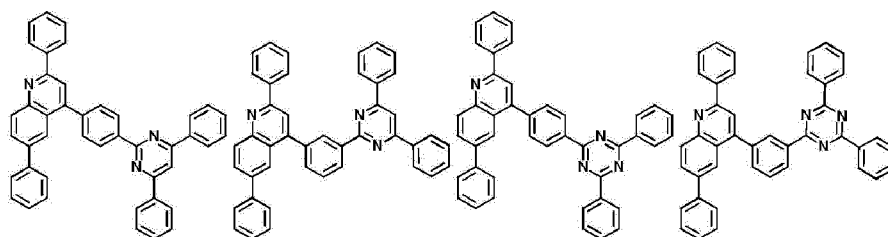
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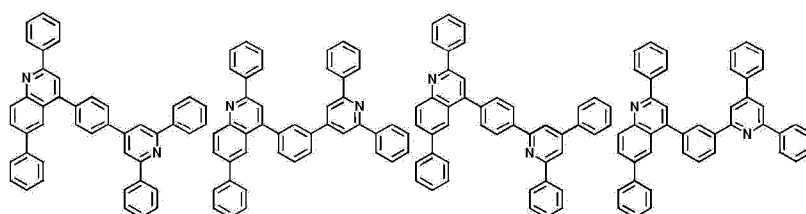
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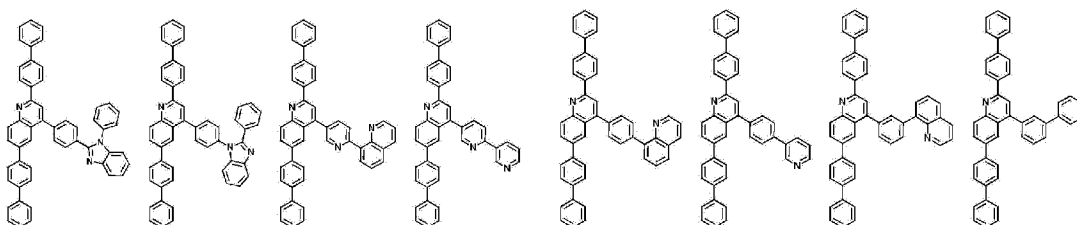
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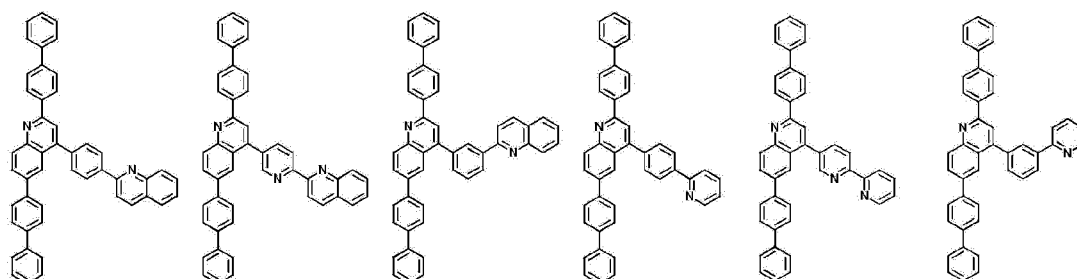
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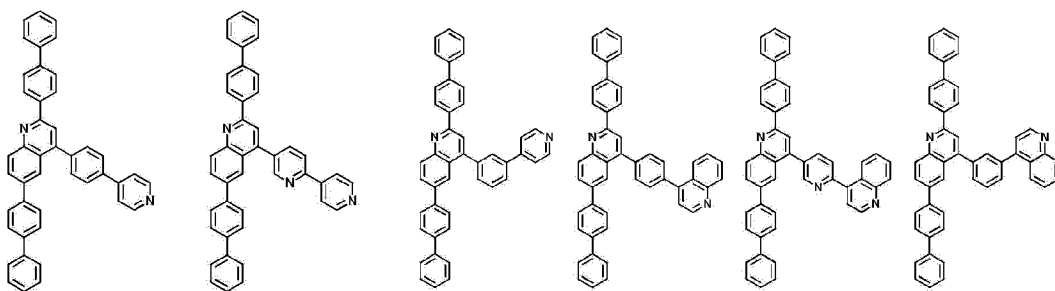
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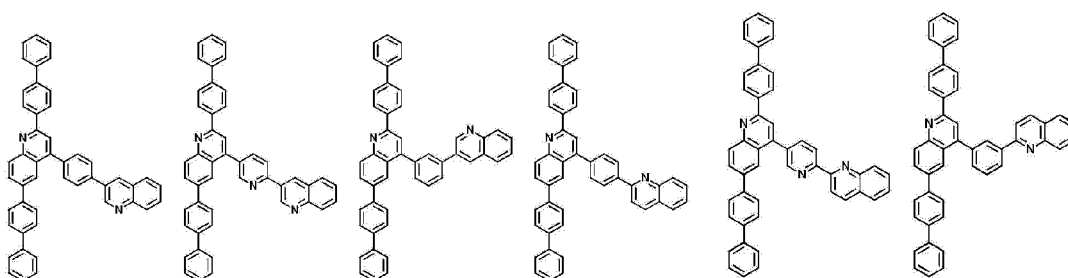
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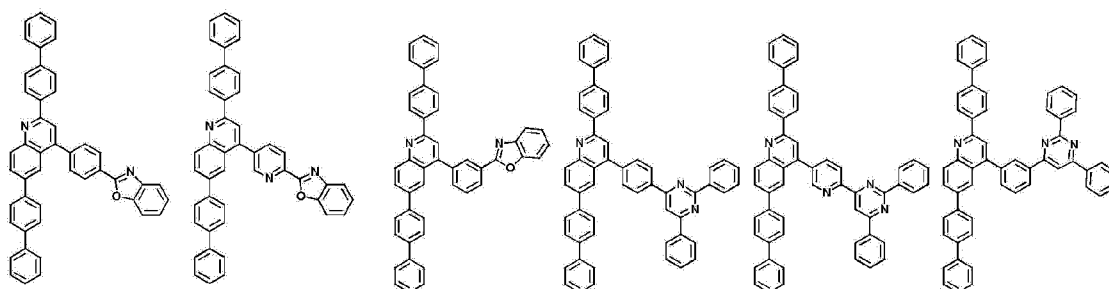


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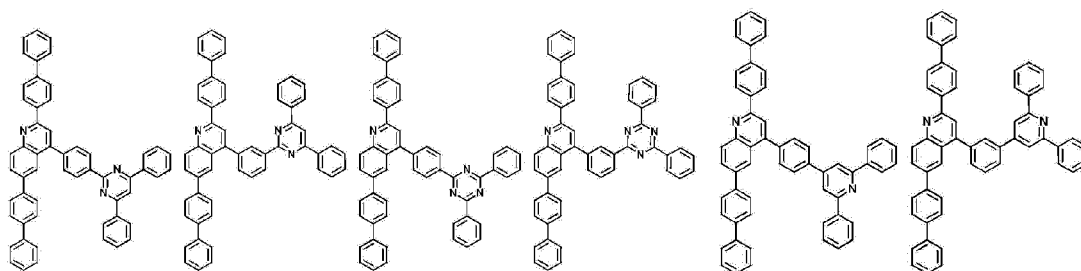


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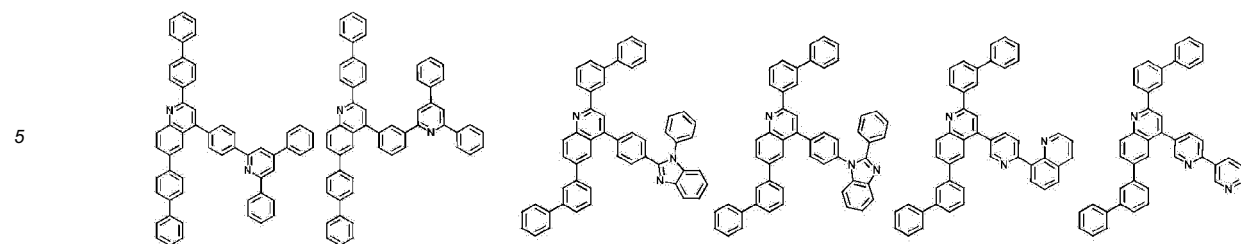
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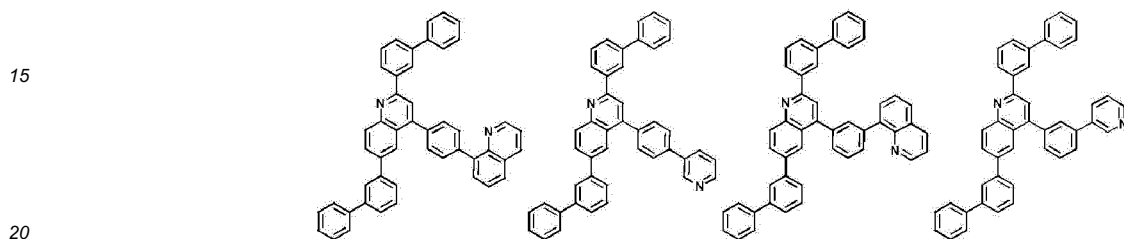


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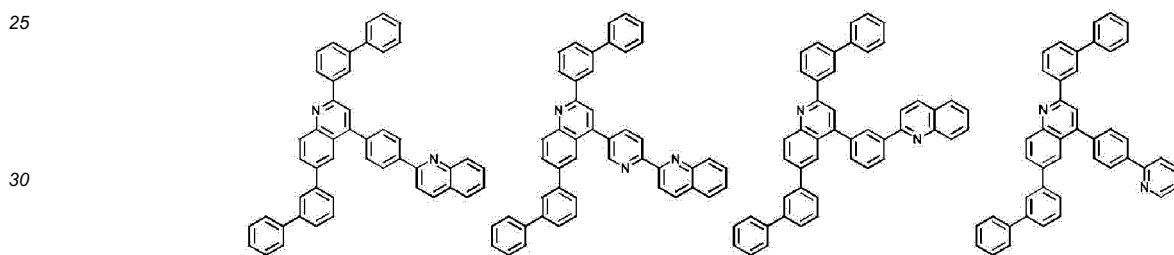
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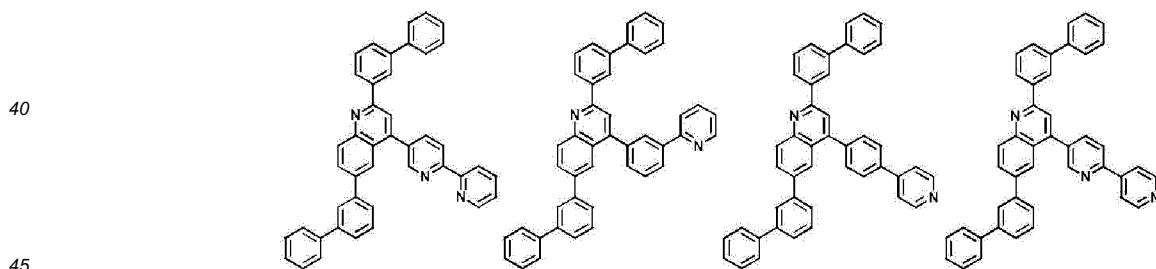
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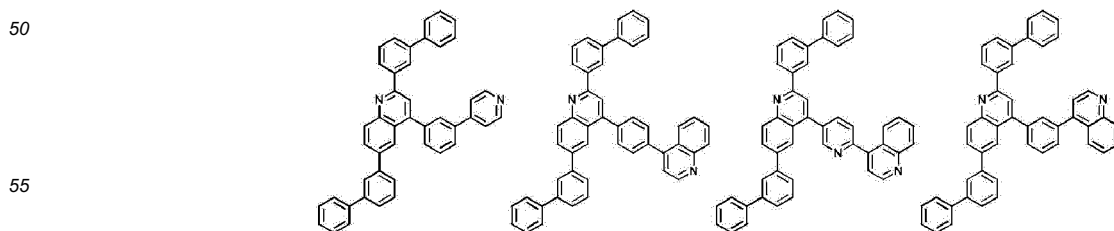
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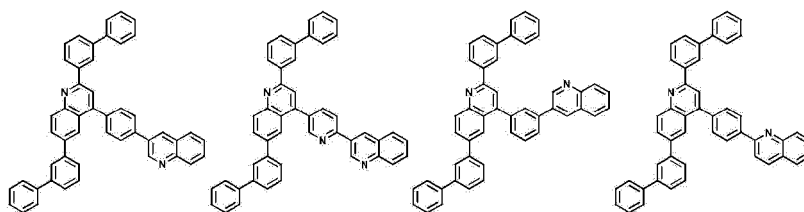
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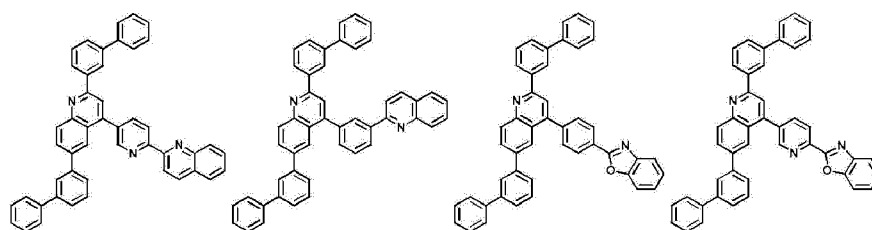
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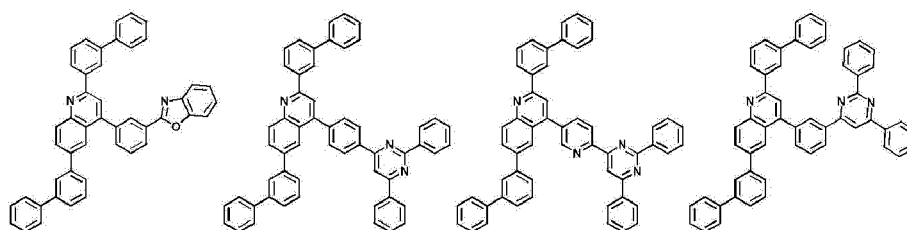
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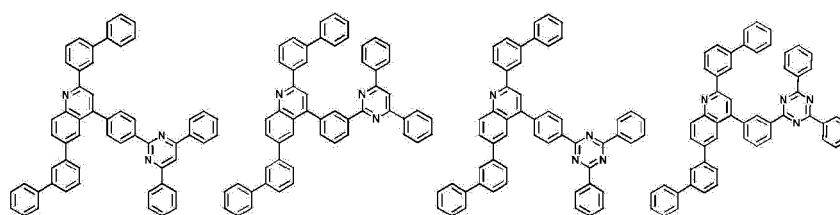
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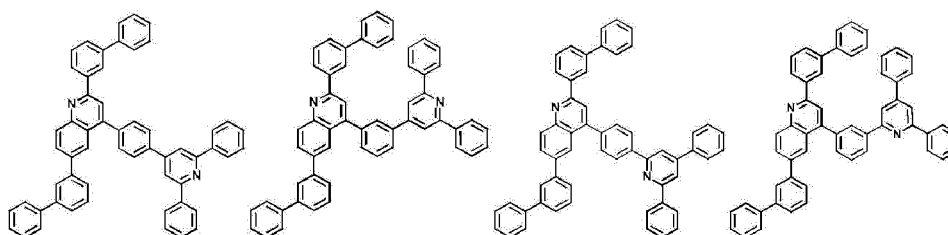


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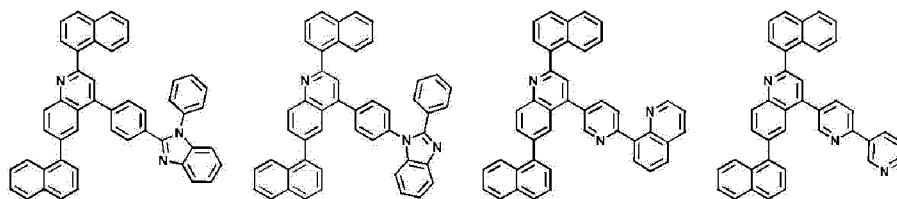
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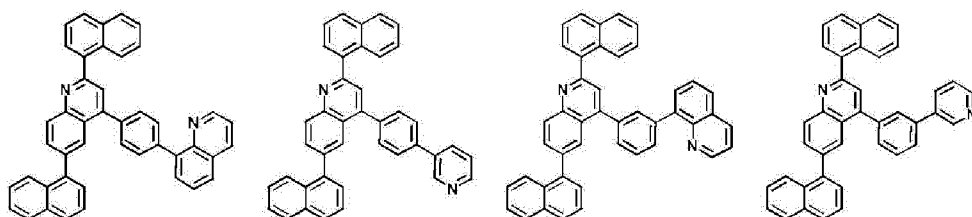
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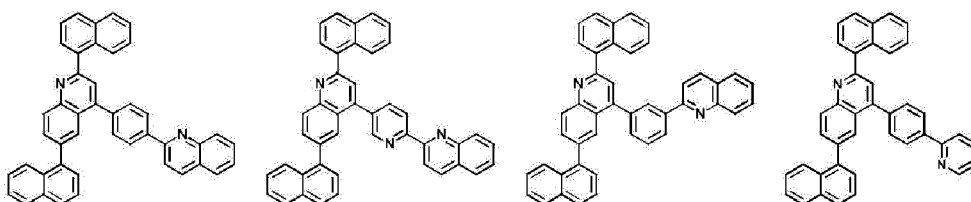
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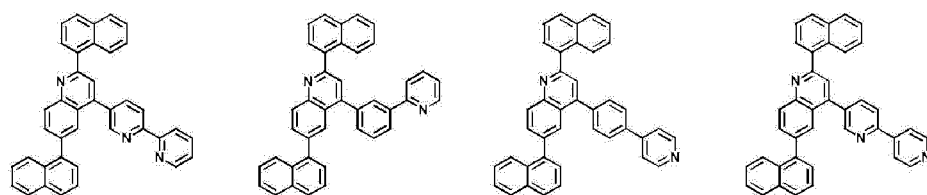
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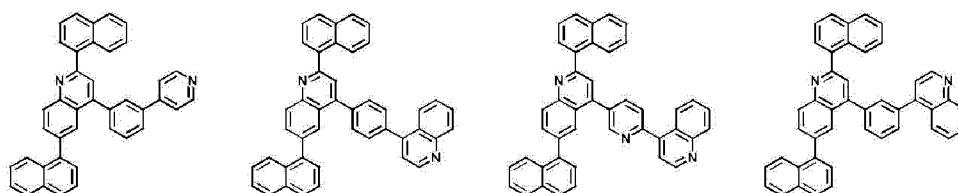
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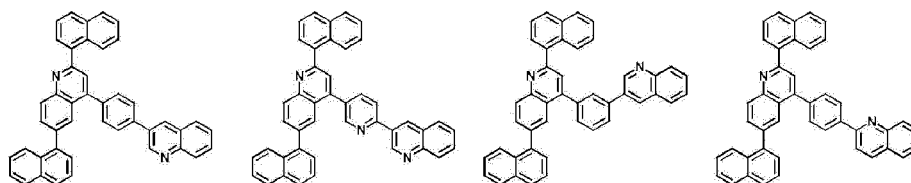
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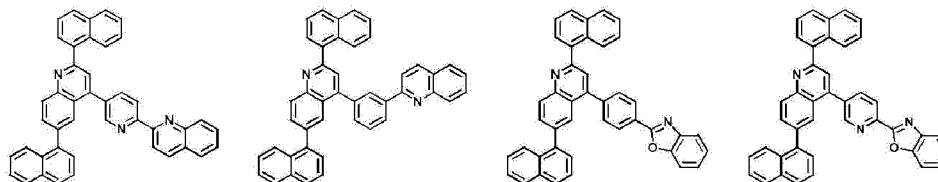
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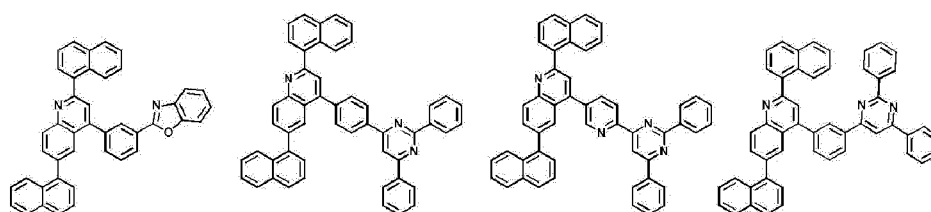
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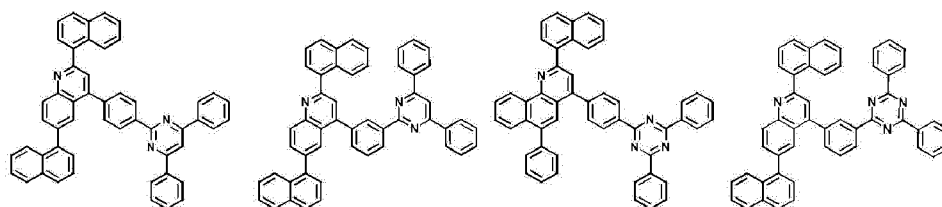
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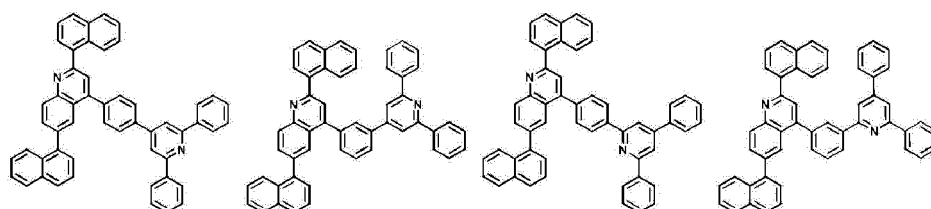
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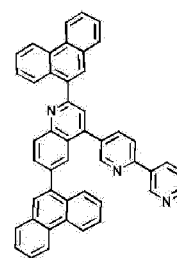
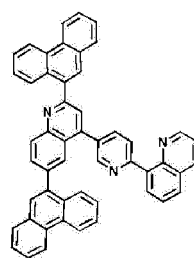
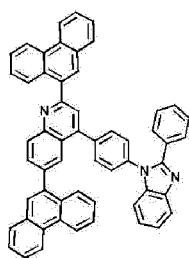
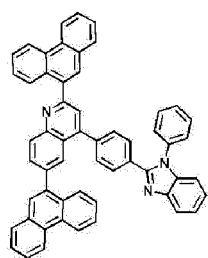
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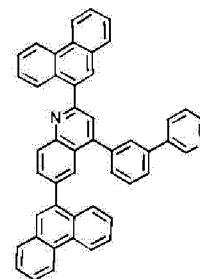
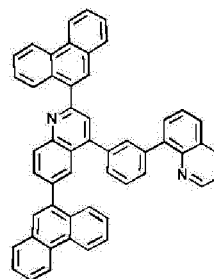
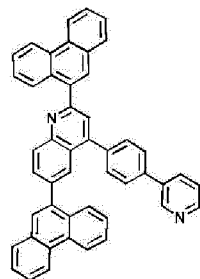
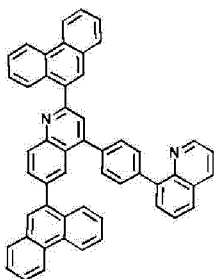
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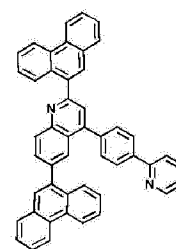
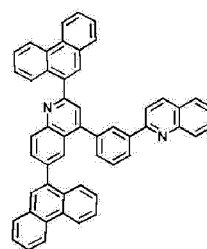
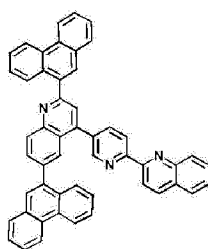
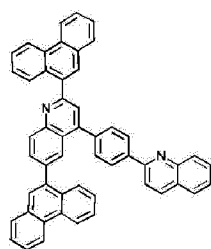
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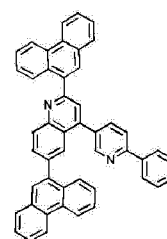
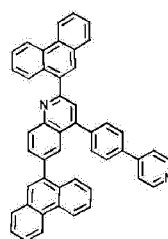
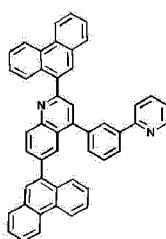
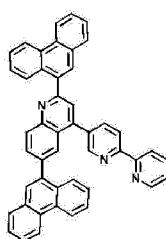
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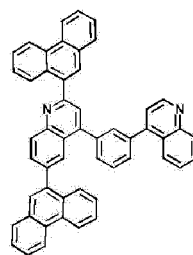
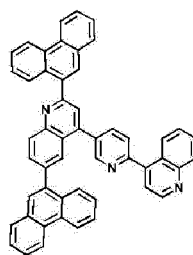
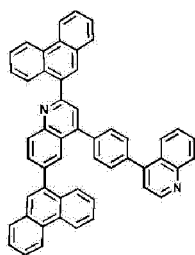
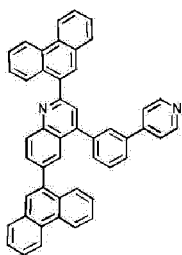


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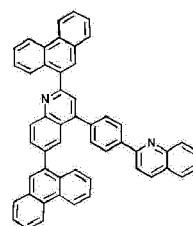
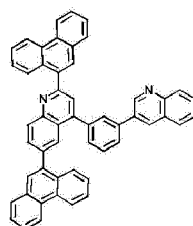
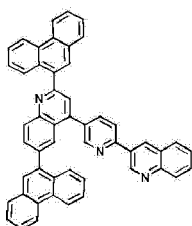
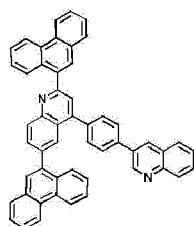
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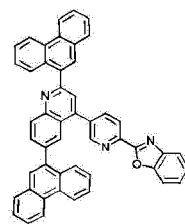
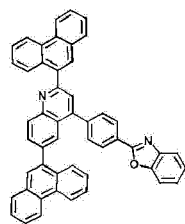
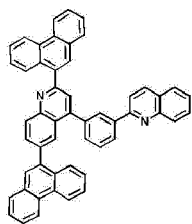
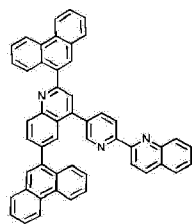
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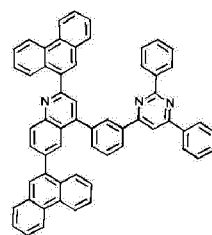
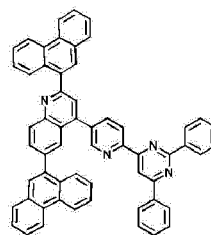
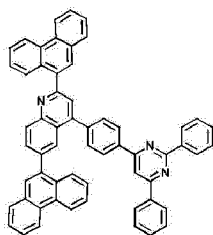
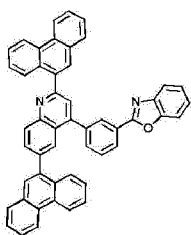
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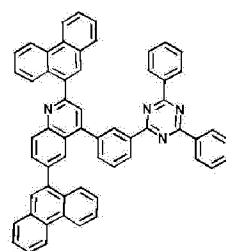
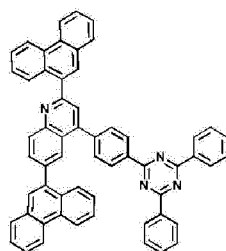
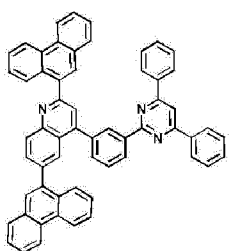
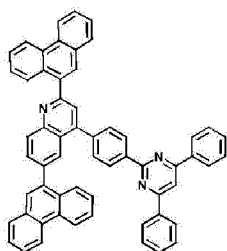


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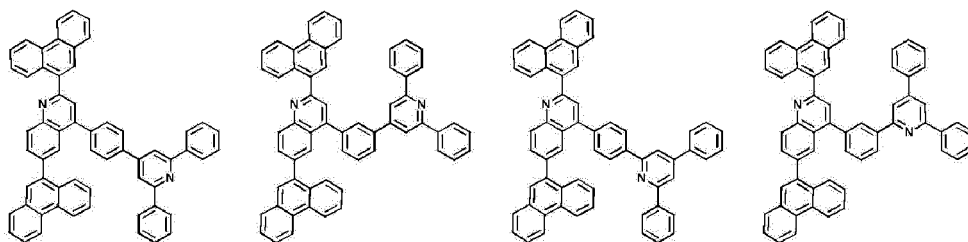


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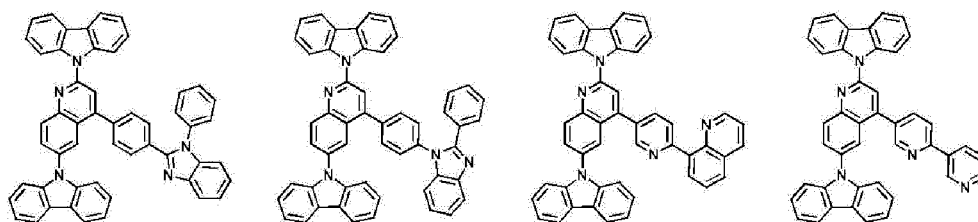
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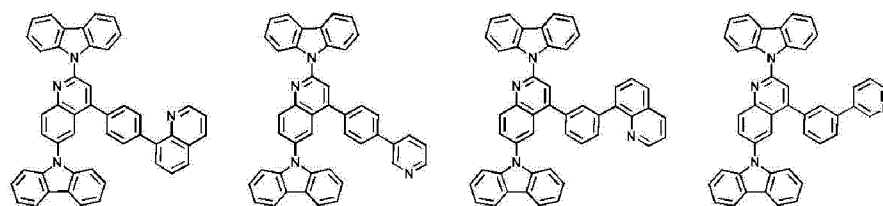
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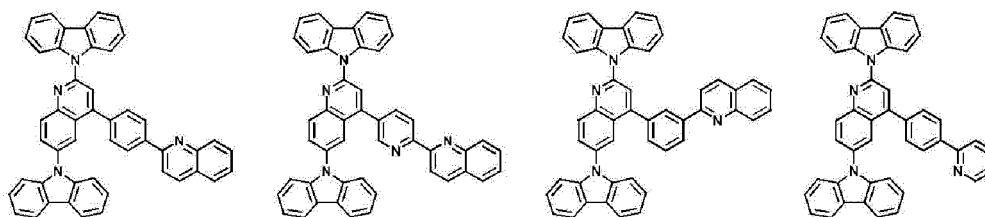


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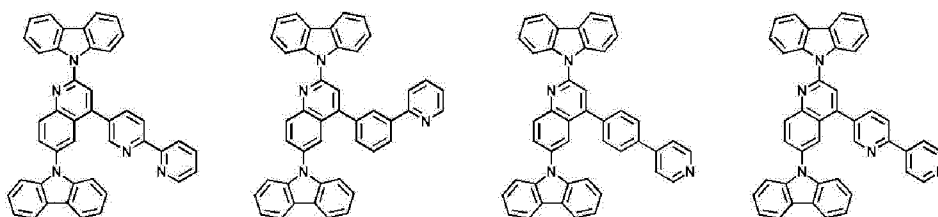
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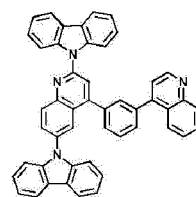
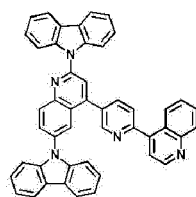
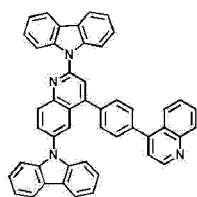
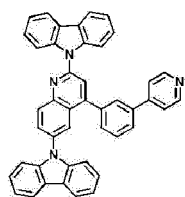
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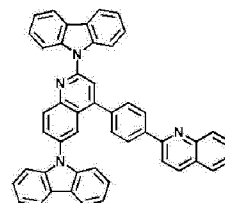
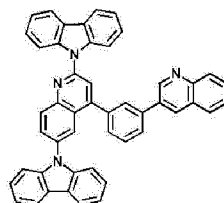
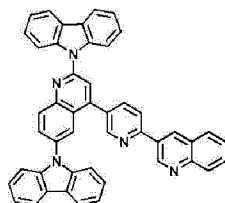
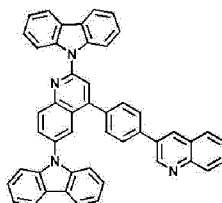
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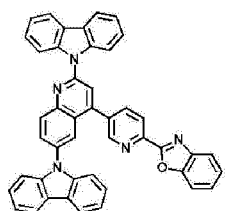
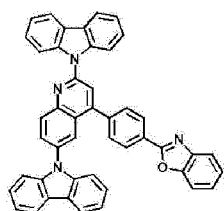
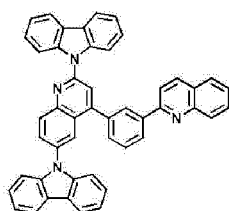
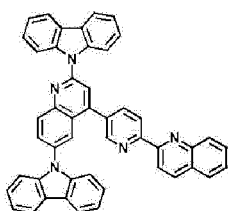
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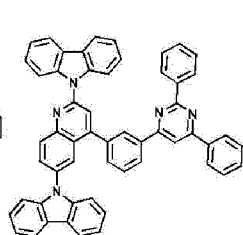
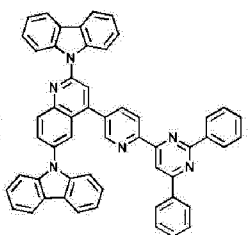
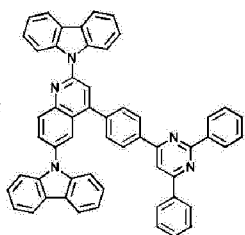
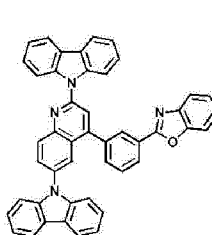
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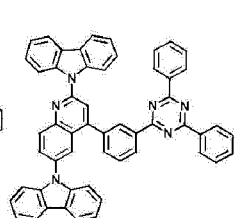
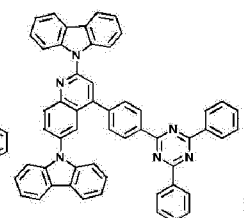
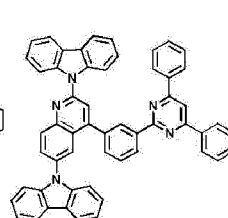
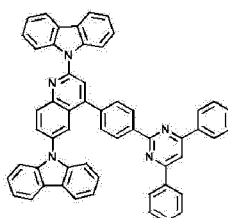


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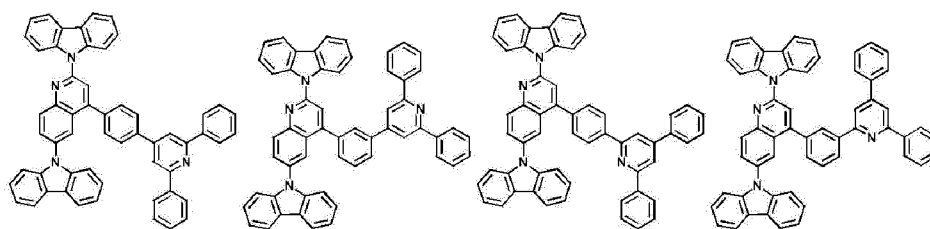
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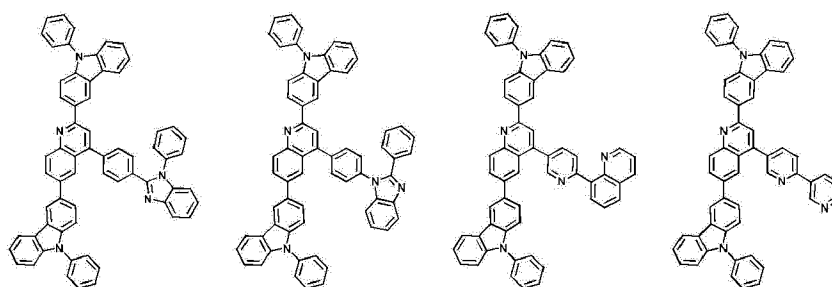


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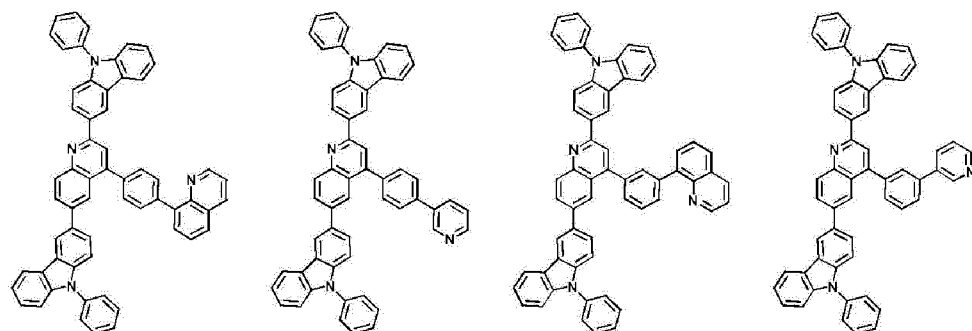
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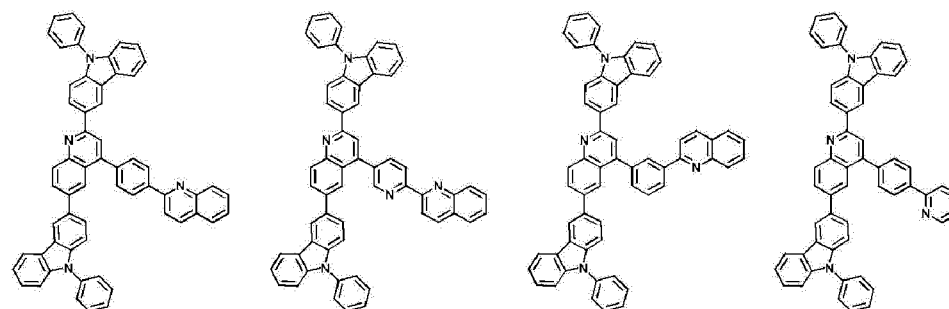
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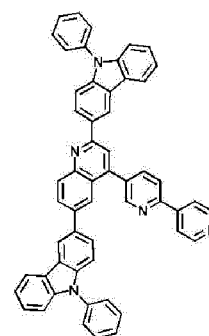
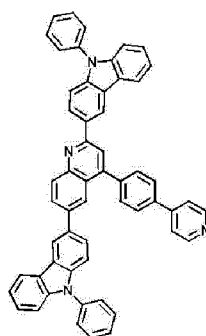
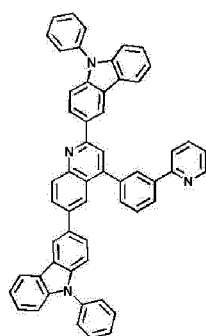
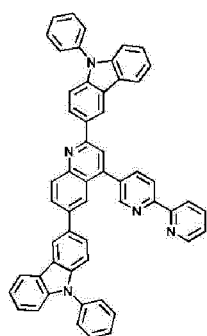


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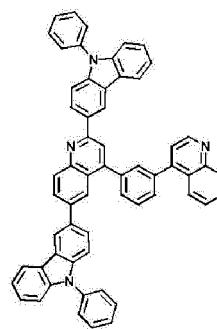
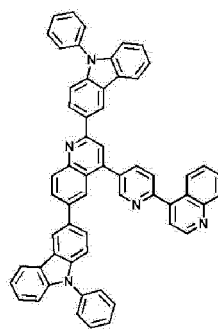
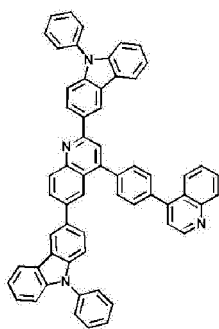
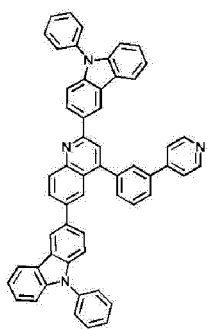
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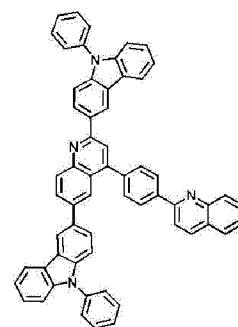
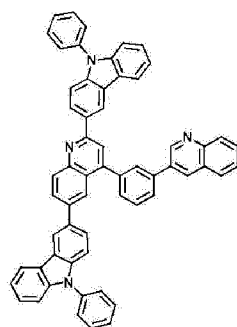
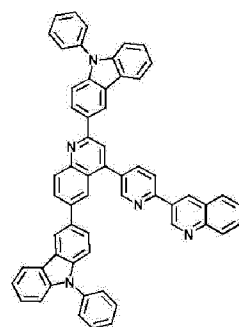
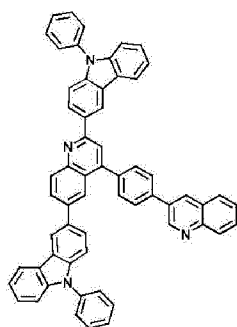


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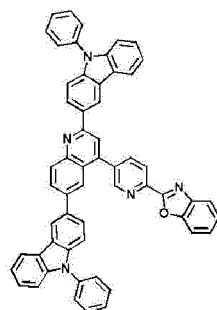
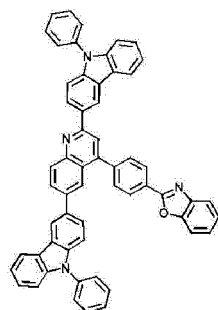
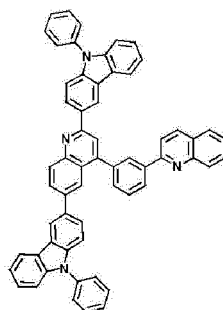
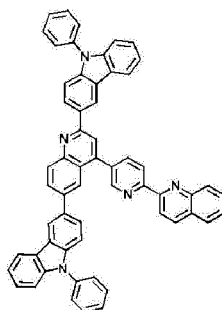


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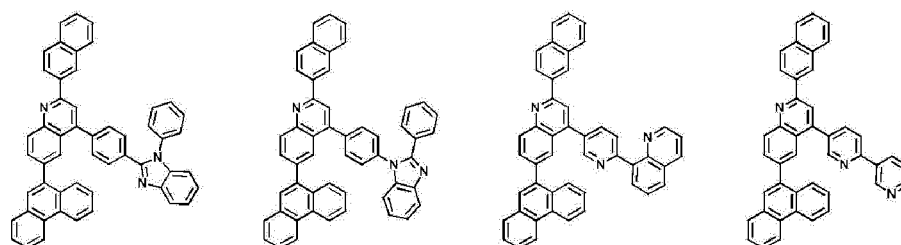
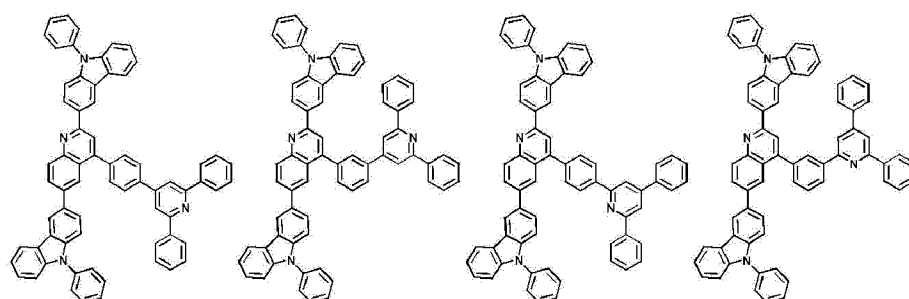
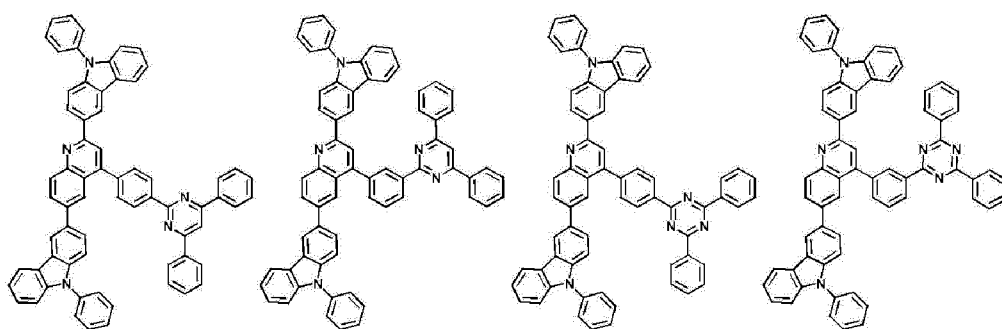
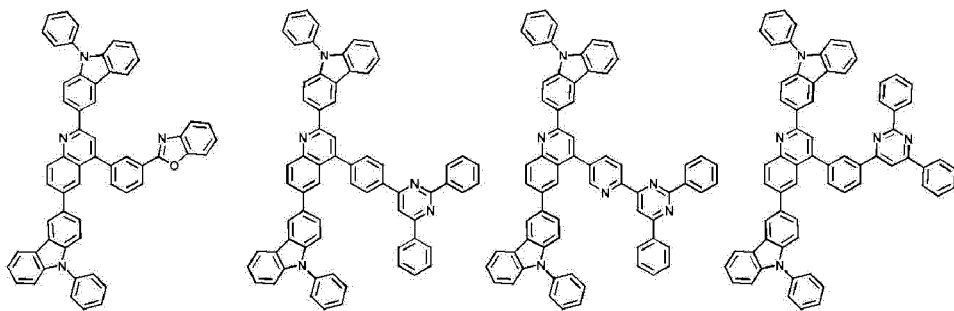
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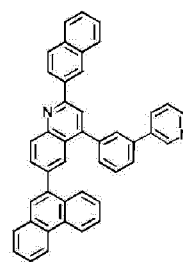
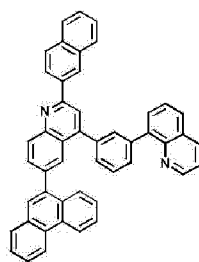
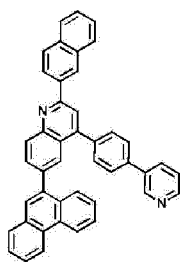
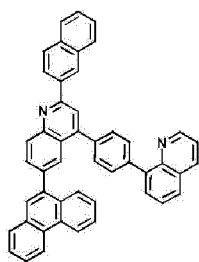
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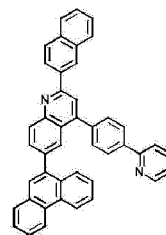
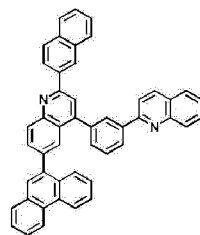
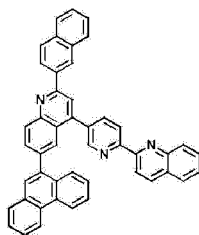
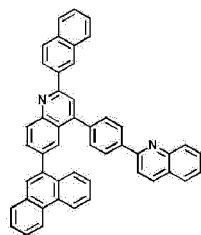
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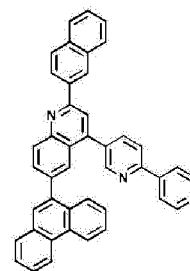
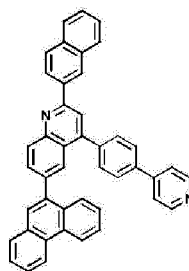
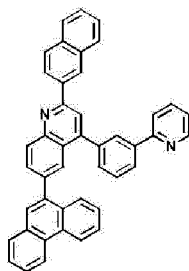
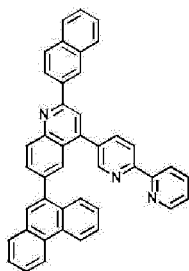
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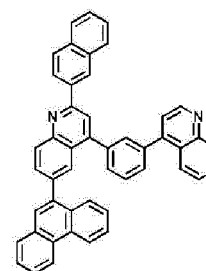
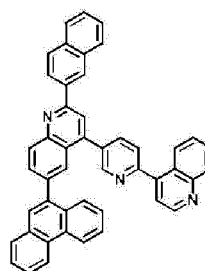
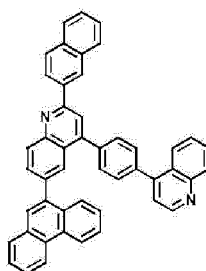
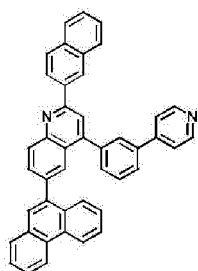


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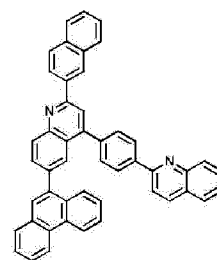
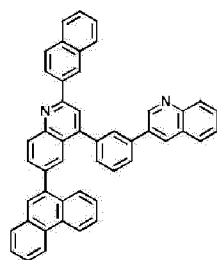
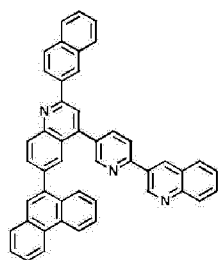
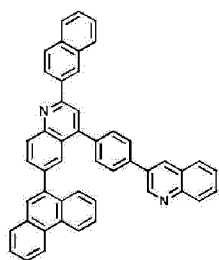


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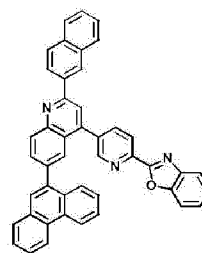
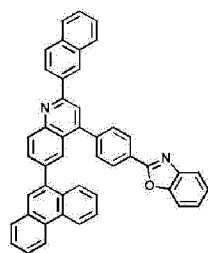
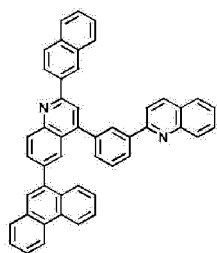
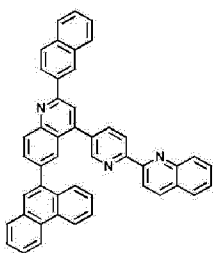
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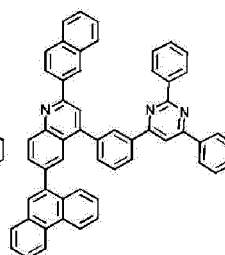
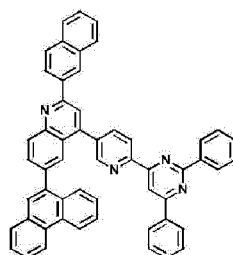
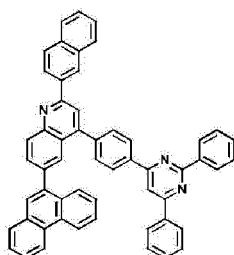
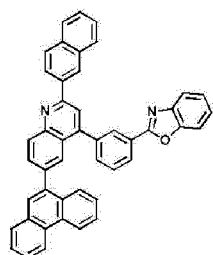
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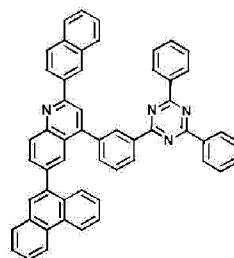
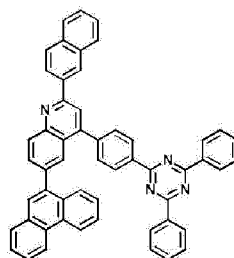
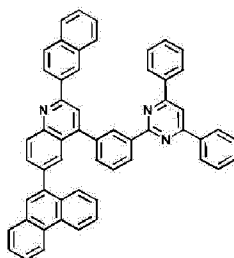
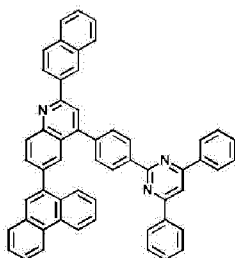
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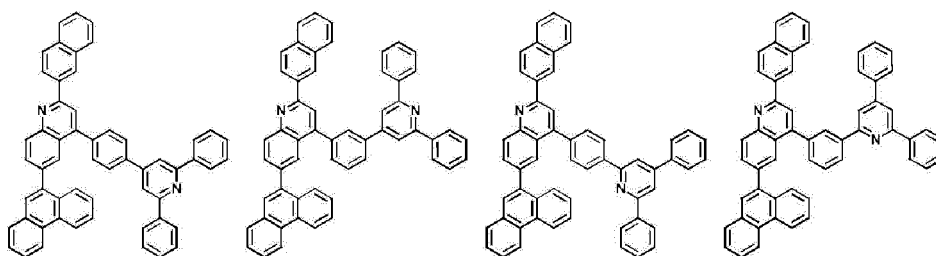
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[A-353] [A-354] [A-355] [A-356]



[A-357] [A-358] [A-359] [A-360]



[0061] In another embodiment of the present invention, provided is an organic optoelectric device that includes an anode, cathode, and an organic thin layer interposed between the anode and the cathode, wherein at least one layer of the organic thin layer includes the compound according to one embodiment of the present invention.

[0062] The compound for an organic optoelectric device is used in an organic thin layer and thus improves life-span characteristics, efficiency characteristic, electrochemical stability and thermal stability of an organic optoelectric device, and lowers a driving voltage.

[0063] The organic thin layer may be specifically an emission layer.

[0064] The organic optoelectric device may be an organic light emitting diode, an organic photoelectric device, an organic solar cell, an organic transistor, an organic photo conductor drum, or an organic memory device.

[0065] More specifically, the organic optoelectric device may be an organic light emitting diode.

[0066] Hereinafter, as an example of an organic optoelectric device, an organic light emitting diode is described referring to drawings.

[0067] FIGS. 1 and 2 are cross-sectional views of an organic light emitting diode according to one embodiment.

[0068] Referring to FIG. 1, an organic optoelectric device 100 according to one embodiment includes an anode 120 and a cathode 110 facing each other and an organic layer 105 between the anode 120 and the cathode 110.

[0069] The anode 120 may be made of a conductor having a high work function to help hole injection, for example metal, a metal oxide and/or a conductive polymer. The anode 120 may include, for example a metal or an alloy thereof such as nickel, platinum, vanadium, chromium, copper, zinc, and gold; metal oxide such as zinc oxide, indium oxide, indium tin oxide (ITO), and indium zinc oxide (IZO); a combination of a metal and an oxide such as ZnO and Al or SnO₂ and Sb; a conductive polymer such as poly (3-methylthiophene), poly (3,4-(ethylene-1,2-dioxy)thiophene) (PEDT), polypyrrole and polyaniline, but is not limited thereto.

[0070] The cathode 110 may be made of a conductor having a low work function to help electron injection, for example a metal, a metal oxide and/or a conductive polymer. The cathode material may include, for example a metal or an alloy thereof such as magnesium, calcium, sodium, potassium, titanium, indium, yttrium, lithium, gadolinium, aluminum silver, tin, lead, cesium, barium, and the like; a multi-layer structured material such as LiF/Al, LiO₂/Al, LiF/Ca, LiF/Al and BaF₂/Ca, but is not limited thereto.

[0071] The organic layer 105 includes an emission layer 130 including the compound.

[0072] The emission layer 130 may include, for example the compound at alone or with at least two of the compounds, or as a mixture with other different compound from the compound. When the compound is mixed with the other compound, for example they may be included as a host and a dopant, wherein the compound may be, for example included as a host. The host may be, for example phosphorescent host or fluorescent host, for example a phosphorescent host.

[0073] When the compound is included as a host, the dopant may be selected from well-known inorganic, organic, organic/inorganic compound as a dopant.

[0074] Referring to FIG. 2, an organic light emitting diode 200 further includes a hole auxiliary layer 140 as well as an emission layer 230. The hole auxiliary layer 140 may further increase hole injection and/or hole mobility between the anode 120 and emission layer 230 and block electrons. The hole auxiliary layer 140 may be, for example a hole transport layer (HTL), a hole injection layer (HIL), and/or an electron blocking layer, and may include at least one layer. The compound may be included in the emission layer 230 and/or the hole auxiliary layer 140.

[0075] Even though not being shown in FIG. 1 or 2, the organic layer 105 may further include an electron injection layer (EIL), an electron transport layer (ETL), an auxiliary electron transport layer (ETL), a hole transport layer (HTL), an auxiliary hole transport layer (HTL), a hole injection layer (HIL), or a combination thereof. The compound of the present invention may be included in these organic layers. The organic light emitting diodes 100 and 200 may be manufactured by forming an anode or a cathode on a substrate, forming an organic layer in accordance with a dry coating method such as evaporation, sputtering, plasma plating, and ion plating; or a wet coating method such as spin coating, slit coating, dipping, flow coating and inkjet printing; and forming a cathode or an anode thereon.

[0076] The organic light emitting diode may be applied to an organic light emitting diode (OLED) display.

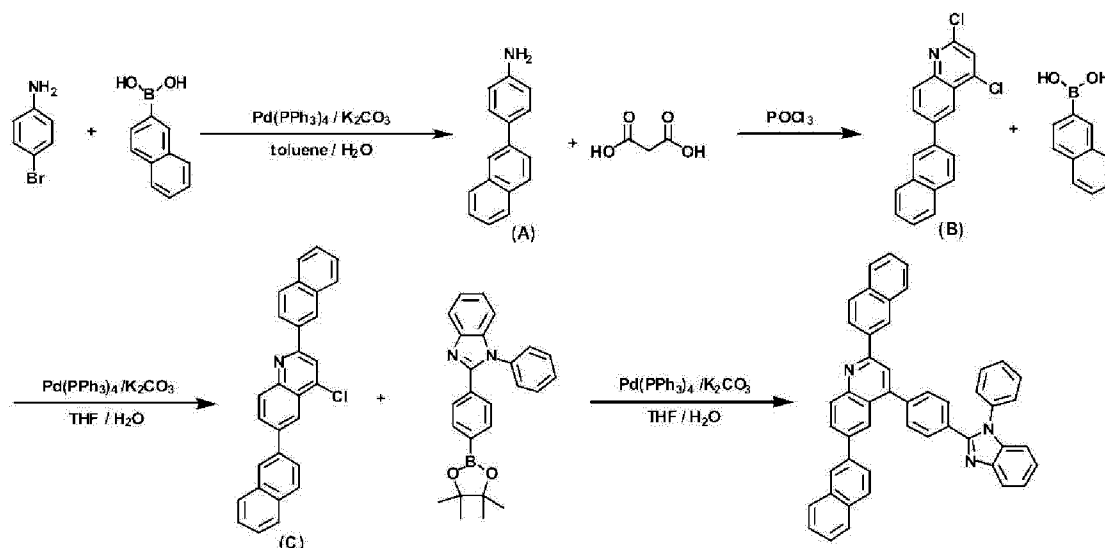
[0077] Hereinafter, the embodiments are illustrated in more detail with reference to examples. These examples, however, are not in any sense to be interpreted as limiting the scope of the invention.

(Preparation of Compound for Organic Optoelectric Device)

Example 1: Synthesis of Compound A-1

[0078] The compound A-1 was synthesized through the following 4 step processes provided in the following Reaction Scheme 1.

[Reaction Scheme 1]



First Step; Synthesis of Intermediate Product (A)

[0079] 58.0 g (337.2 mmol) of 4-bromoaniline, 69.6 g (404.6 mmol) of 2-naphthaleneboronic acid and 9.7 g (8.4 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 1700 ml of a toluene solvent, a solution obtained by dissolving 93.2 g (674.3 mmol) of potassium carbonate (K_2CO_3) in 600 ml of water was added thereto, and the mixture was reacted at 100 °C for 12 hours. Then, an aqueous layer was removed from the obtained reactant, and a product obtained after removing a solvent under a reduced pressure was washed with water and methanol. The obtained solid mixture was separated through column and dried, obtaining 50.0 g of a yellow solid intermediate product (A) (a yield: 68 %).

Second step; Synthesis of Intermediate Product (B)

[0080] 27.0 g (123.1 mmol) of the intermediate product A and 19.2 g (184.7 mmol) of malonic acid were dissolved in 110 ml of a phosphorus oxychloride ($POCl_3$) solvent, and the solution was reacted at 140 °C for 4 hours.

[0081] The obtained reactant was poured into ice water, and a solid formed therein was filtered and washed with water and a sodium bicarbonate-saturated aqueous solution. The obtained solid mixture was washed with methanol and dried, obtaining 9.6 g of a light yellow solid intermediate product (B) (a yield: 24 %).

Third step: Synthesis of Intermediate Product (C)

[0082] 9.5 g (29.3 mmol) of the intermediate product (B), 5.0 g (29.3 mmol) of 2-naphthaleneboronic acid and 1.0 g (0.9 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 190 ml of tetrahydrofuran (THF), a solution obtained by dissolving 8.1 g (58.6 mmol) of potassium carbonate (K_2CO_3) in 95 ml of water was added thereto, and the mixture was reacted at 70 °C for 12 hours. The resultant was cooled down to room temperature, and a solid mixture formed therein was filtered through a glass filter. The residue was recrystallized with monochloro benzene, and a crystal precipitated therein was separated through a filter and washed with monochloro benzene and dried, obtaining 9.19 g of a white solid intermediate product (C) (a yield: 75 %).

Fourth Step: Synthesis of Compound A-1

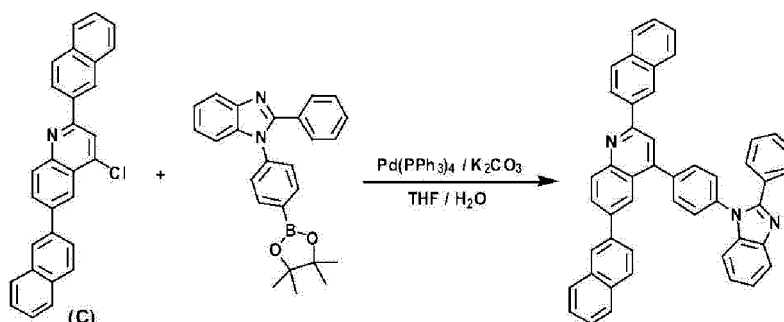
[0083] 9.0 g (21.6 mmol) of the intermediate product (C), 10.3 g (23.9 mmol) of 1-phenyl-2-(4-(4,4,5,5-tetramethyl-

1,3,2-dioxaborolan-2-yl)phenyl)-1H-benzimidazole and 0.8 g (0.7 mmol) of tetrakis(triphenylphosphine)palladium [Pd(PPh₃)₄] were dissolved in 180 ml of tetrahydrofuran (THF), a solution obtained by dissolving 11.9 g (86.6 mmol) of potassium carbonate (K₂CO₃) in 90 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a produce obtained therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate obtained therefrom was separated and clean with toluene and then, dried, obtaining 11.0 g of a white solid compound (a yield: 81 %). (calculation value : 649.78, measurement value : MS[M+1] 650.08)

Example 2: Synthesis of Compound A-2

[0084] The compound A-2 was synthesized through the following Reaction Scheme 2.

[Reaction Scheme 2]

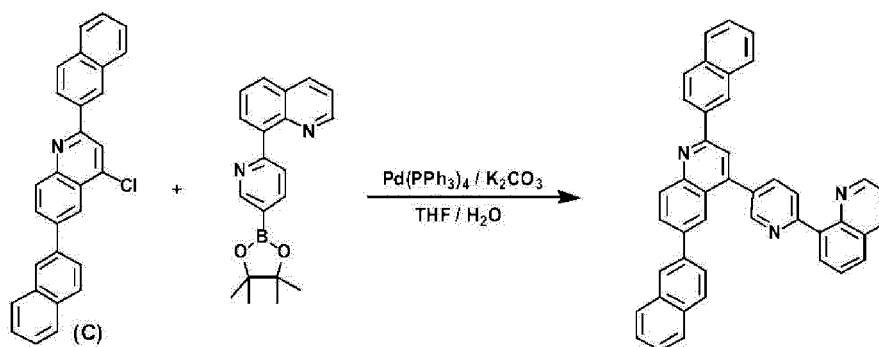


[0085] 15.0 g (36.1 mmol) of the intermediate product (C), 17.2 g (43.3 mmol) of 2-phenyl-1-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-1H-benzimidazole and 1.3 g (1.1 mmol) of tetrakis(triphenylphosphine)palladium [Pd(PPh₃)₄] were dissolved in 300 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 19.9 g (144.3 mmol) of potassium carbonate (K₂CO₃) in 150 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a produce therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate obtained therefrom was separated through a filter and washed with toluene and then, dried, obtaining 18.0 g of a white solid compound (a yield: 77%). (calculation value : 649.78, measurement value : MS[M+1] 650.08)

Example 3: Synthesis of Compound A-3

[0086] The compound A-3 was synthesized through the following Reaction Scheme 3.

[Reaction Scheme 3]



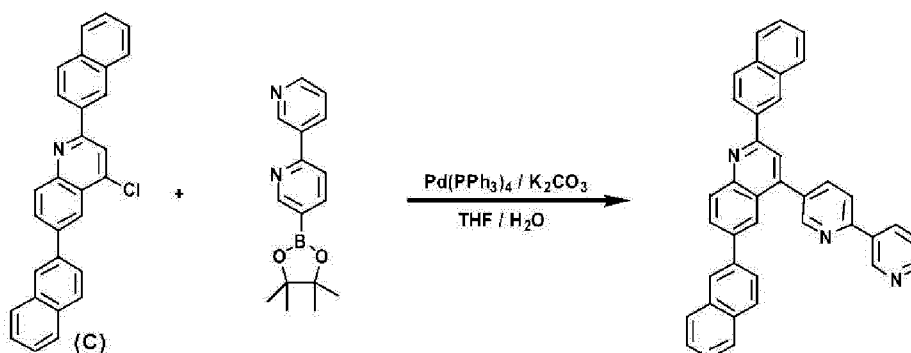
[0087] 12.0 g (28.9 mmol) of the intermediate product (C), 11.5 g (34.6 mmol) of 8-(5-(4,4,5,5-tetramethyl-1,3,2-

dioxaborolan-2-yl)-pyridine-2-yl)quinoline and 1.0 g (0.9 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 240 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 15.9 g (115.4 mmol) of potassium carbonate (K_2CO_3) in 120 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 13.0 g of a white solid compound (a yield: 77 %). (calculation value: 585.69, measurement value: $MS[M+1]$ 585.99)

Example 4: Synthesis of Compound A-4

[0088] The compound A-4 was synthesized through the following Reaction Scheme 4.

[Reaction Scheme 4]

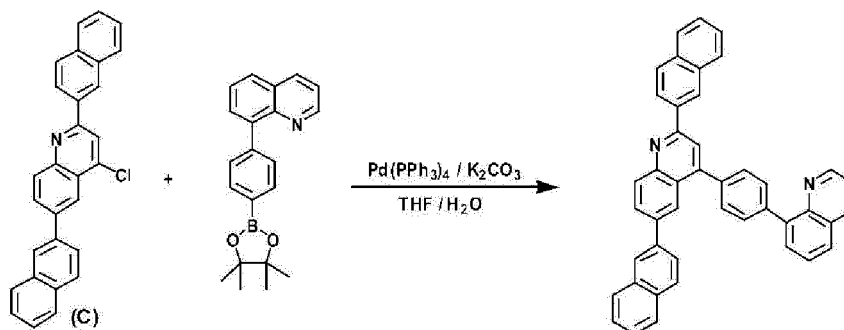


[0089] 11.5 g (27.7 mmol) of the intermediate product (C), 9.4 g (33.2 mmol) of 3-(5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-pyridine-2-yl)pyridine and 1.0 g (0.8 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 230 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 15.3 g (110.6 mmol) of potassium carbonate (K_2CO_3) in 115 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 9.7 g of a white solid compound (a yield: 65 %). (calculation value : 535.64, measurement value : $MS[M+1]$ 535.94)

Example 5: Synthesis of Compound A-5

[0090] The compound A-5 was synthesized through the following Reaction Scheme 5.

[Reaction Scheme 5]



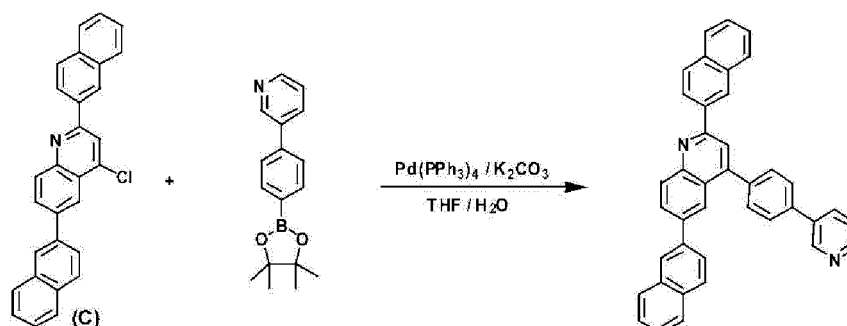
[0091] 15.0 g (36.1 mmol) of the intermediate product (C), 14.3 g (43.3 mmol) of 8-(4-(4,4,5,5-tetramethyl-1,3,2-

dioxaborolan-2-yl)-phenyl)quinoline and 1.3 g (1.1 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 300 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 19.9 g (144.3 mmol) of potassium carbonate (K_2CO_3) in 150 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 15.0 g of a white solid compound (a yield: 71%). (calculation value : 584.71, measurement value : $MS[M+1]$ 585.01)

Example 6: Synthesis of Compound A-6

[0092] The compound A-6 was synthesized through the following Reaction Scheme 6.

[Reaction Scheme 6]

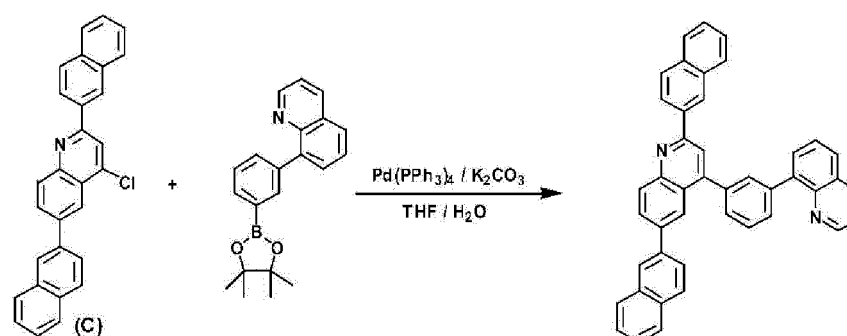


[0093] 16.0 g (34.3 mmol) of the intermediate product (C), 11.6 g (41.2 mmol) of 3-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)pyridine and 1.2 g (1.0 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 320 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 19.0 g (56.5 mmol) of potassium carbonate (K_2CO_3) in 160 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 15.0 g of a white solid compound (a yield: 82 %). (calculation value: 534.65, measurement value: $MS[M+1]$ 534.95)

Example 7: Synthesis of Compound A-7

[0094] The compound A-7 was synthesized through the following Reaction Scheme 7.

[Reaction Scheme 7]



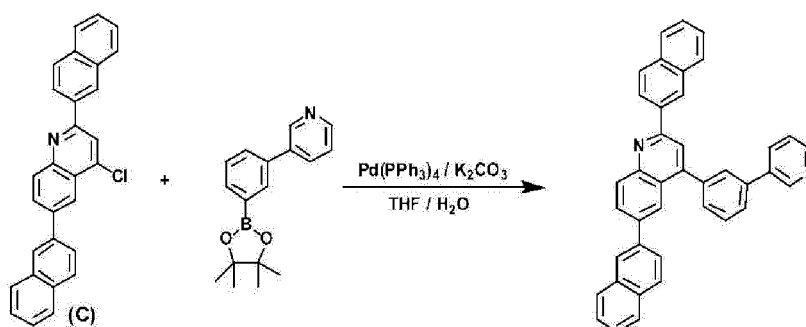
[0095] 16.0 g (38.5 mmol) of the intermediate product (C), 15.3 g (46.2 mmol) of 8-(3-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)quinolin-2-yl)quinoline and 1.3 g (1.2 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dis-

solved in 320 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 21.3 g (153.9 mmol) of potassium carbonate (K_2CO_3) in 160 ml of water, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 15.8 g of a white solid compound (a yield: 70 %). (calculation value: 584.71, measurement value: $MS[M+1]$ 585.01)

Example 8: Synthesis of Compound A-8

[0096] The compound A-8 was synthesized through the following Reaction Scheme 8.

[Reaction Scheme 8]

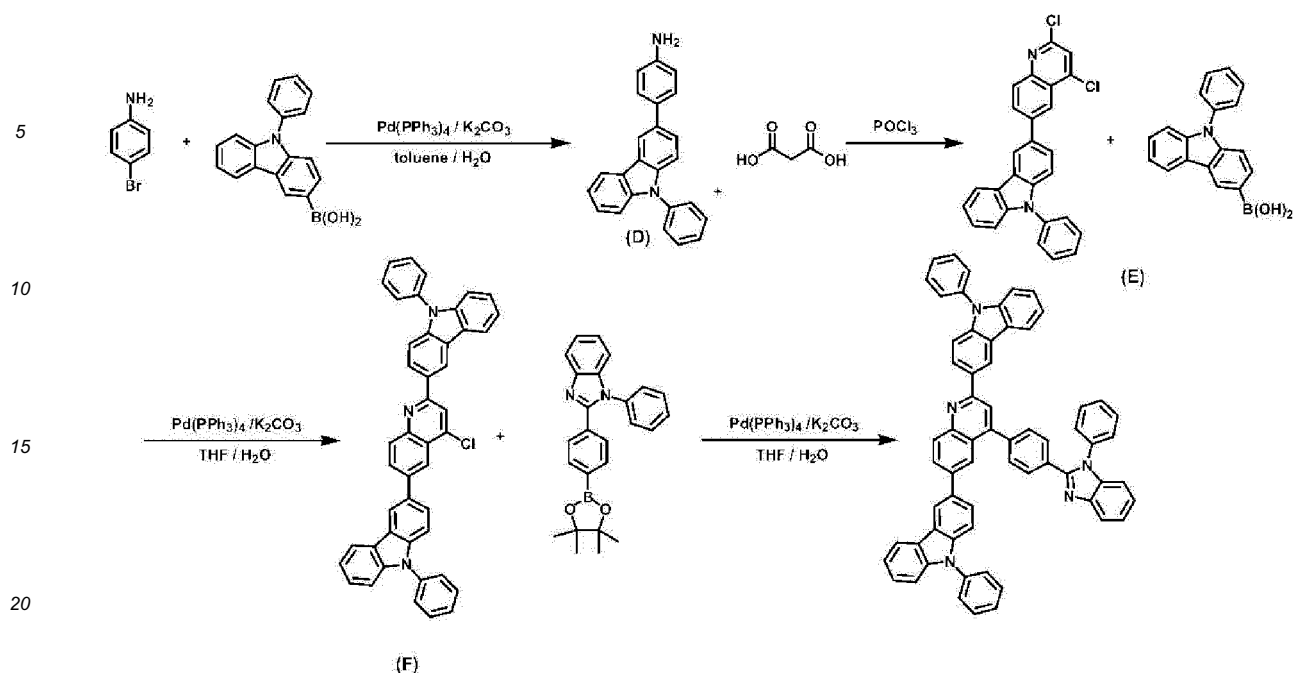


[0097] 10.0 g (21.5 mmol) of the intermediate product (C), 7.2 g (25.8 mmol) of 3-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)pyridine and 0.7 g (0.6 mmol) of tetrakis(triphenylphosphine)palladium $[Pd(PPh_3)_4]$ were dissolved in 200 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 11.9 g (85.8 mmol) of potassium carbonate (K_2CO_3) in 100 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 9.5 g of a white solid compound (a yield: 83 %). (calculation value : 534.65, measurement value : $MS[M+1]$ 534.95)

Example 9: Synthesis of Compound A-281

[0098] The compound A-281 was synthesized through 4 steps provided in the following Reaction Scheme 9.

[Reaction Scheme 9]



First Step; Synthesis of Intermediate Product (D)

[0099] 70.0 g (406.9 mmol) of 4-bromoaniline, 114.5 g (398.8 mmol) of (9-phenyl-9H-carbazol-3-yl)boronic acid and 11.8 g (10.2 mmol) of tetrakis(triphenylphosphine)palladium $[\text{Pd(PPh}_3)_4]$ were dissolved in 1050 ml of toluene as a solvent, a solvent obtained by dissolving 112.5 g (813.9 mmol) of potassium carbonate (K_2CO_3) in 490 ml of water was added thereto, and the mixture was reacted at 100 °C for 12 hours. Then, an aqueous layer was removed from the reactant, a solvent was removed therefrom under a reduced pressure, and a product therefrom was washed with water and methanol. The obtained solid mixture was separated through column and dried, obtaining 90.0 g of a yellow solid intermediate product (D) (a yield: 66 %).

Second Step; Synthesis of Intermediate Product (E)

[0100] 60.0 g (179.4 mmol) of the intermediate product (D) and 28.0 g (269.1 mmol) of malonic acid were dissolved in 164 ml of phosphorus oxychloride (POCl_3) as a solvent, and the solution was reacted at 140 °C for 4 hours. The obtained reactant was poured into ice water, and a solid formed therein was filtered and washed with water and sodium bicarbonate-saturated aqueous solution. The obtained solid mixture was washed with methanol and dried, obtaining 28.0 g of a light yellow solid intermediate product (E) (a yield: 35 %).

Third Step; Synthesis of Intermediate Product (F)

[0101] 20.0 g (45.5 mmol) of the intermediate product (E), 13.1 g (45.5 mmol) of (9-phenyl-9H-carbazol-3-yl)boronic acid and 1.6 g (1.4 mmol) of tetrakis(triphenylphosphine)palladium $[\text{Pd(PPh}_3)_4]$ were dissolved in 400 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 12.6 g (91.1 mmol) of potassium carbonate (K_2CO_3) in 200 ml of water was added thereto, and the mixture was reacted at 70 °C for 12 hours. The resultant was cooled down to room temperature, and a solid mixture formed therein was filtered through a glass filter. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 23.0 g of a white solid intermediate product (F) (a yield: 78 %).

Fourth Step; Synthesis of Compound A-281

[0102] 15.0 g (23.2 mmol) of the intermediate product (F), 11.0 g (27.9 mmol) of 1-phenyl-2-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-1H-benzimidazole and 0.8 g (0.7 mmol) of tetrakis(triphenylphosphine)palladium were dissolved in 300 ml of tetrahydrofuran (THF) as a solvent, a solution obtained by dissolving 12.8 g (92.9 mmol) of potassium carbonate (K_2CO_3) was dissolved in 150 ml of water was added thereto, and the mixture was reacted at 90 °C for 12 hours. Then, a solvent was removed from the obtained reactant under a reduced pressure, and a product

therefrom was washed with water and methanol. The residue was recrystallized with toluene, and a precipitate therefrom was separated through a filter and washed with toluene and then, dried, obtaining 16.0 g of a white solid compound (a yield: 78 %). (calculation value: 880.04, measurement value: MS[M+1] 880.34)

5 (Manufacture of Organic Light Emitting Diode)

Example 10

[0103] As for an anode, 1000 Å-thick ITO was used, and as for a cathode, 1000 Å-thick aluminum (Al) was used.

[0104] Specifically, illustrating a method of manufacturing an organic light emitting diode, the anode is manufactured by cutting an ITO glass substrate having 15 Ω/cm² of sheet resistance into a size of 50 mm × 50 mm × 0.7 mm, ultrasonic wave-cleaning them in acetone, isopropylalcohol, and pure water for 5 minutes respectively, and UV ozone cleaning them for 30 minutes.

[0105] On the glass substrate, a 65 nm--thick hole injection layer (HIL) was formed by depositing N1,N1'-(biphenyl-4,4'-diyl)bis(N1-(naphthalen-2-yl)-N4,N4-diphenylbenzene-1,4-diamine), and then subsequently a 40 nm--thick hole transport layer (HTL) was formed by depositing N,N'-di(1-naphthyl)-N,N'-diphenylbenzidine.

[0106] A 25 nm-thick emission layer was formed by depositing 4 % of N,N,N',N'-tetrakis (3,4-dimethylphenyl)chrysene-6,12-diamine and 96 % of 9-(3-(naphthalen-1-yl)phenyl)-10-(naphthalene-2-yl)anthracene.

[0107] Subsequently, a 30 nm-thick electron transport layer (ETL) was formed by depositing the compound prepared in Example 1.

[0108] On the electron transport layer (ETL), an Liq/Al electrode was formed by vacuum-depositing Liq in a thickness of 0.5 nm as an electron injection layer (EIL), and then vacuum-depositing Al in a thickness of 100 nm.

Example 11

[0109] An organic light emitting diode was manufactured according to the same method as Example 10 except for using the compound of Example 3 instead of the compound of Example 1 as an electron transport layer (ETL).

Example 12

[0110] An organic light emitting diode was manufactured according to the same method as Example 10 except for using the compound of Example 1 and Liq at a 1:1 ratio instead of the compound of Example 1 as an electron transport layer (ETL).

Example 13

[0111] An organic light emitting diode was manufactured according to the same method as Example 10 except for using the compound of Example 3 and Liq at a 1:1 ratio instead of the compound of Example 1 as an electron transport layer (ETL).

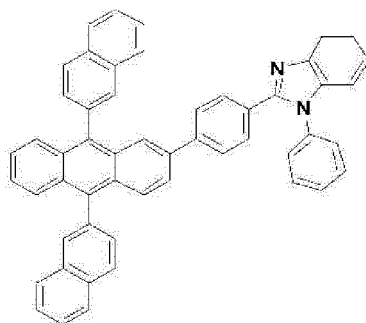
Example 14

[0112] An organic light emitting diode was manufactured according to the same method as Example 10 except for using the compound of Example 4 and Liq at a 1:1 ratio instead of the compound of Example 1 as an electron transport layer (ETL).

Comparative Example 1

[0113] An organic light emitting diode was manufactured according to the same method as Example 10 except for using the compound of Chemical Formula R-1 and Liq at a 1:1 ratio instead of the compound of Example 1 as an electron transport layer (ETL).

[Chemical Formula R-1]

**Comparative Example 2**

[0114] An organic light emitting diode was manufactured according to the same method as Example 13 except for using the compound of Chemical Formula R-1 and Liq at a 1:1 ratio instead of the compound of Example 1 as an electron transport layer (ETL).

(Performance Measurement of Organic Light Emitting Diode)

[0115] Current density and luminance changes depending on a voltage and luminous efficiency of each organic light emitting diode according to Examples 10 to 14 and Comparative Examples 1 and 2 were measured. The measurements were specifically performed in the following method, and the results were provided in the following Table 1.

1) Measurement of Current Density Change Depending on Voltage Change

[0116] Current values flowing in the unit device of the organic light emitting diodes according to Examples 10 to 14, Comparative Examples 1 and 2 were measured for, while increasing the voltage using a current-voltage meter (Keithley 2400), and the measured current values were divided by an area to provide the results.

2) Measurement of Luminance Change Depending on Voltage Change

[0117] Luminance of the organic light emitting diodes according to Examples 10 to 14, Comparative Examples 1 and 2 was measured for luminance, while increasing the voltage using a luminance meter (Minolta Cs-1000A).

3) Measurement of Luminous Efficiency and Power Efficiency

[0118] Current efficiency (cd/A) at the same current density (10 mA/cm²) were calculated by using the luminance, current density, and voltages (V) from the items 1) Measurement of Current Density Change Depending on Voltage Change and 2) Measurement of Luminance Change Depending on Voltage Change, and the results are shown in Table 1.

4) Color Coordinate

[0119] The color coordinate of each organic light emitting diode according to Examples 10 to 14 and Comparative Examples 1 and 2 was measured at 6000 cd/m² by using a luminance meter (Keithley 2635B).

(Table 1)

	Luminance 500 cd/m ²				
	Driving voltage (V)	Luminous efficiency (cd/A)	Power efficiency (lm/W)	CIE	
				x	y
Example 10	5.1	4.3	2.6	0.14	0.06
Example 11	4.4	4.6	3.3	0.14	0.05

(continued)

	Luminance 500 cd/m ²				
	Driving voltage (V)	Luminous efficiency (cd/A)	Power efficiency (lm/W)	CIE	
				x	y
Comparative Example 1	5.1	3.7	2.3	0.14	0.05
Example 12	3.8	6.2	5.1	0.14	0.05
Example 13	3.9	5.3	4.3	0.14	0.05
Example 14	3.8	6.0	5.0	0.15	0.06
Comparative Example 2	4.2	5.4	4.1	0.14	0.05

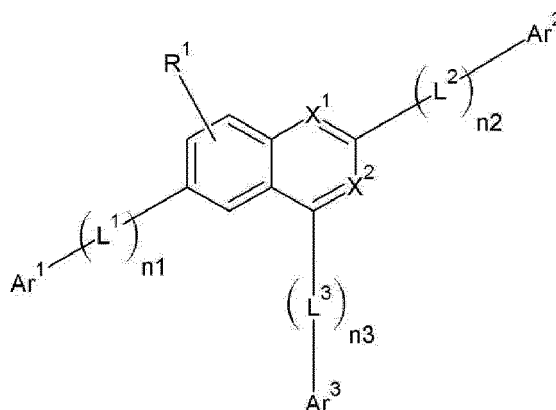
[0120] As shown in Table 1, the organic light emitting diodes according to Examples 10 to 14 showed improved characteristics in terms of a driving voltage, luminous efficiency and/or power efficiency compared with the organic light emitting diodes according to Comparative Examples 1 and 2.

[0121] While this invention has been described in connection with what is presently considered to be practical exemplary embodiments, it is to be understood that the invention is not limited to the disclosed embodiments, but, on the contrary, is intended to cover various modifications and equivalent arrangements included within the spirit and scope of the appended claims. Therefore, the aforementioned embodiments should be understood to be exemplary but not limiting the present invention in any way.

Claims

1. A compound for an organic optoelectric device represented by Chemical Formula 1:

[Chemical Formula 1]



wherein, in Chemical Formula 1,

X¹ and X² are each independently N or CR['],

at least one of X¹ and X² is N,

Ar³ is a substituted or unsubstituted C2 to C30 heteroaryl group,

Ar¹ and Ar² are each independently a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C2 to C30 heteroaryl group,

L¹ to L³ are each independently a substituted or unsubstituted C6 to C30 arylene group or a substituted or unsubstituted C2 to C30 heteroarylene group,

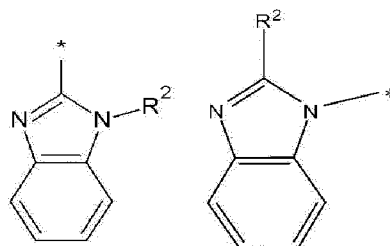
n₁ to n₃ are independently 0 or 1, and

R¹ and R['] are each independently hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group,

a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

2. The compound for an organic optoelectric device of claim 1, wherein the Ar³ is a substituent represented by Chemical Formula 2 or Chemical Formula A:

[Chemical Formula 2] [Chemical Formula A]

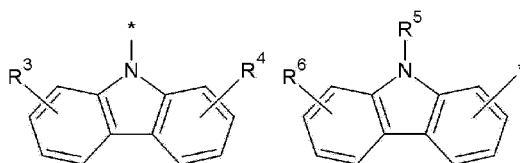


wherein, in Chemical Formulae 2 and A,

R² is hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

3. The compound for an organic optoelectric device of claim 1, wherein the Ar¹ and Ar² are each independently a substituted or unsubstituted C6 to C30 aryl group.
4. The compound for an organic optoelectric device of claim 1, wherein the Ar¹ and Ar² are a substituted or unsubstituted phenyl group, a substituted or unsubstituted biphenyl group, a substituted or unsubstituted naphthyl group, a substituted or unsubstituted anthracenyl group, or a substituted or unsubstituted phenanthrenyl group.
5. The compound for an organic optoelectric device of claim 1, wherein at least one of the Ar¹ and Ar² is a substituent represented by Chemical Formula 3 or 4:

[Chemical Formula 3] [Chemical Formula 4]



wherein, in Chemical Formulae 3 and 4,

R³ to R⁶ are independently hydrogen, deuterium, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, a substituted or unsubstituted C2 to C30 heteroaryl group or a substituted or unsubstituted silyl group.

6. The compound for an organic optoelectric device of claim 1, wherein the L³ is a substituted or unsubstituted C2 to C30 heteroarylene group.
7. The compound for an organic optoelectric device of claim 1, wherein the L³ is a substituted or unsubstituted pyridinylene group, a substituted or unsubstituted pyrimidinylene group, or a substituted or unsubstituted triazinylene group.
8. The compound for an organic optoelectric device of claim 1, wherein the Ar³ is a substituted or unsubstituted pyridinyl group, a substituted or unsubstituted pyrimidinyl group, a substituted or unsubstituted triazinyl group, a substituted or unsubstituted quinolinyl group or a substituted or unsubstituted isoquinolinyl group.

9. The compound for an organic optoelectric device of claim 1, wherein the X¹ is N, and the X² is CR'.

10. An organic light emitting diode comprising

5 an anode, a cathode and
 at least one organic thin layer between the anode and the cathode,
 wherein at least one layer of the organic thin layer comprises the compound of any one of claim 1 to claim 9.

10 11. The organic light emitting diode of claim 10, wherein the organic thin layer comprises an electron injection layer (EIL), an electron transport layer (ETL), a hole injection layer (HIL), a hole transport layer (HTL), or an emission layer.

12. The organic light emitting diode of claim 10, wherein the organic thin layer is an electron injection layer (EIL) or an electron transport layer (ETL).

15 13. The organic light emitting diode of claim 10, wherein the organic thin layer is an emission layer.

14. The organic light emitting diode of claim 10, wherein the compound is used as a host in an emission layer.

20 15. A display device comprising the organic light emitting diode of claim 10.

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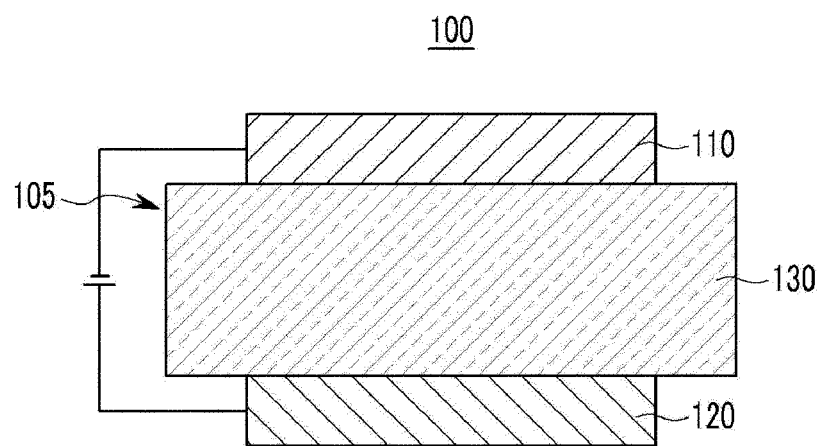
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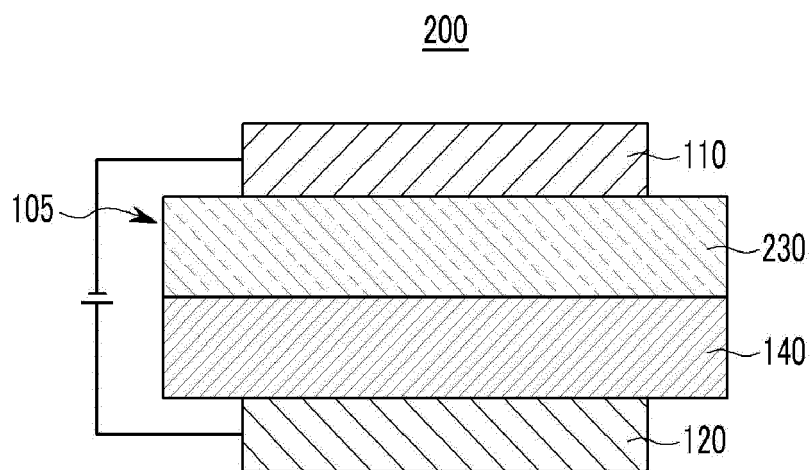
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【Figure 1】





INTERNATIONAL SEARCH REPORT

International application No.
PCT/KR2013/007247

A. CLASSIFICATION OF SUBJECT MATTER

C09K 11/06(2006.01)i, C07D 401/10(2006.01)i, C07D 401/14(2006.01)i, H01L 51/50(2006.01)i

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C09K 11/06; H01L 51/54; C07D 215/04; C07D 417/04; C07D 215/38; C07F 15/00; C07D 401/14; C07D 401/10; H01L 51/50

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
Korean Utility models and applications for Utility models: IPC as above
Japanese Utility models and applications for Utility models: IPC as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
eKOMPASS (KIPO internal) & Keywords: quinolin, benzimidazole, pyridine, organic electroluminescent device

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 2011-014039 A1 (ROHM AND HAAS ELECTRONIC MATERIALS KOREA LTD.) 03 February 2011 See the entire document, <41>-<74>, especially compound no. 55.	1-15
X	CN102703059 A (JILIN OPTICAL AND ELECTRONIC MATERIALS CO.,LTD.) 03 October 2012 See structural formula (1) and [0021].	1-15
A	WO 01-42218 A1 (KABUSHIKI KAISHA TOYOTA CHUO KENKYUSHO) 14 June 2001 See the claims.	1-15
E	WO 2013-162148 A1 (CHEIL INDUSTRIES INC.) 31 October 2013 See chemical formula 2 and [41]-[325].	1-15

☐ Further documents are listed in the continuation of Box C. ☒ See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

27 FEBRUARY 2014 (27.02.2014)

Date of mailing of the international search report

28 FEBRUARY 2014 (28.02.2014)

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Facsimile No. 82-42-472-7140

Authorized officer

Telephone No.

INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.

PCT/KR2013/007247

Patent document cited in search report	Publication date	Patent family member	Publication date
WO 2011-014039 A1	03/02/2011	CN 102625806 A KR 10-2011-0013220 A US 2012-0217485 A1	01/08/2012 09/02/2011 30/08/2012
CN 102703059 A	03/10/2012	NONE	
WO 01-42218 A1	14/06/2001	JP 04-048780B2 US 2003-0012980 A1 US 6730418 B2	20/02/2008 16/01/2003 04/05/2004
WO 2013-162148 A1	31/10/2013	KR 10-2013-0119740 A	01/11/2013

Form PCT/ISA/210 (patent family annex) (July 2009)

专利名称(译)	用于有机光电器件的化合物，包括其的有机发光二极管，包括有机发光二极管的显示器件		
公开(公告)号	EP2998380A4	公开(公告)日	2016-12-07
申请号	EP2013884807	申请日	2013-08-12
[标]申请(专利权)人(译)	第一毛织株式会社		
申请(专利权)人(译)	第一毛织INC.		
当前申请(专利权)人(译)	Cheil Industries Inc.		
[标]发明人	KANG DONG MIN KIM HYUN JUNG SHIN CHANG JU WON JONG WOO LEE NAM HEON JEONG SOO YOUNG JEONG WOO SEOK JUNG HO KUK CHAE MI YOUNG		
发明人	KANG, DONG-MIN KIM, HYUN-JUNG SHIN, CHANG-JU WON, JONG-WOO LEE, NAM-HEON JEONG, SOO-YOUNG JEONG, WOO-SEOK JUNG, HO-KUK CHAE, MI-YOUNG		
IPC分类号	C09K11/06 C07D401/10 C07D401/14 H01L51/50 C07D215/12 C07D403/10 C07D403/14		
CPC分类号	H01L51/0072 C07D215/12 C07D401/10 C07D401/14 C07D403/10 C07D403/14 C09K11/025 C09K11/06 C09K2211/1007 C09K2211/1011 C09K2211/1029 C09K2211/1033 C09K2211/1044 C09K2211/1059 H01L27/3244 H01L51/0052 H01L51/0058 H01L51/0067 H01L51/0071 H01L51/5012 H01L51/5072 H01L51/5092		
优先权	1020130054001 2013-05-13 KR		
其他公开文献	EP2998380A1		
外部链接	Espacenet		

摘要(译)

本发明提供由化学式1表示的有机光电装置用化合物，包含该化合物的有机发光二极管和包括该有机发光二极管的显示装置。由化学式1表示的有机光电装置用化合物的结构在说明书中描述。用于有机光电装置的化合物提供具有优异的电化学和热稳定性和改善的寿命特性以及在低驱动电压下的高发光效率的有机发光二极管，并且用于有机光电装置的化合物可适用于溶液处理。

